

Supplementary Information

Platinum and Other Metals Doped-Manganese Tungstate as a Novel Catalytic Support for NaBH₄ Hydrolysis and Hydrogen Evolution Under Mild Conditions

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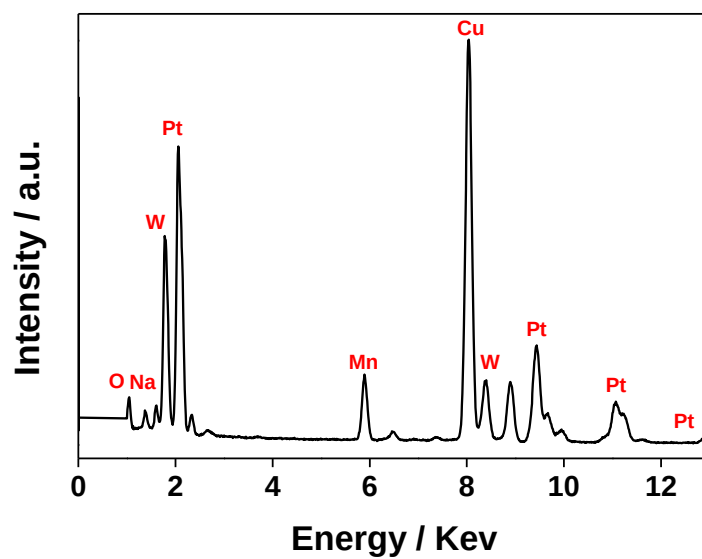


Figure S1. Energy dispersive X-ray spectroscopy (EDS) of Pt NPs/MnWO₄.

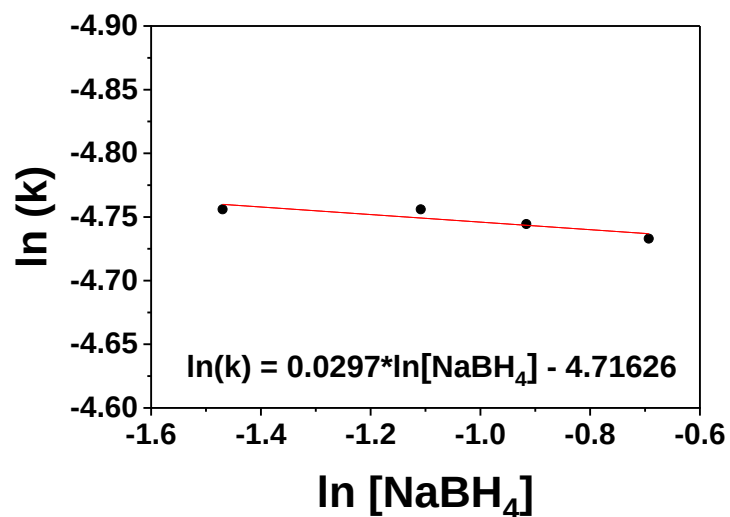


Figure S2. $\ln k$ vs. $\ln [\text{NaBH}_4]$. Experimental conditions: 50 mg of support, dose of 0.025 mmol Pt (0.0049 g), 1.0 mL NaBH₄ (0.230, 0.330, 0.400 and 0.500 mol L⁻¹), constant agitation, and temperature of 293.15 K.

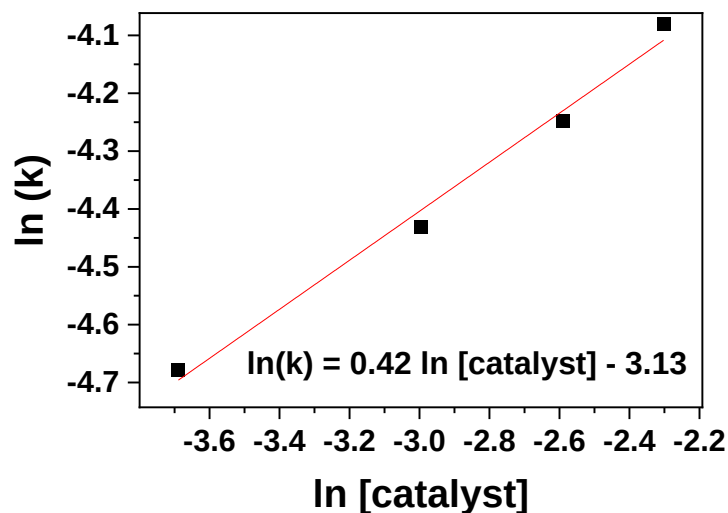


Figure S3. $\ln k$ vs. $\ln$ of the catalyst dose of Pt NPs/MnWO₄. Experimental conditions: 50 mg of support, dose of Pt (0.025, 0.050, 0.075, 0.100 mmol), 1.0 mL NaBH₄ (0.500 mol L⁻¹), constant agitation, and temperature of 293.15 K.

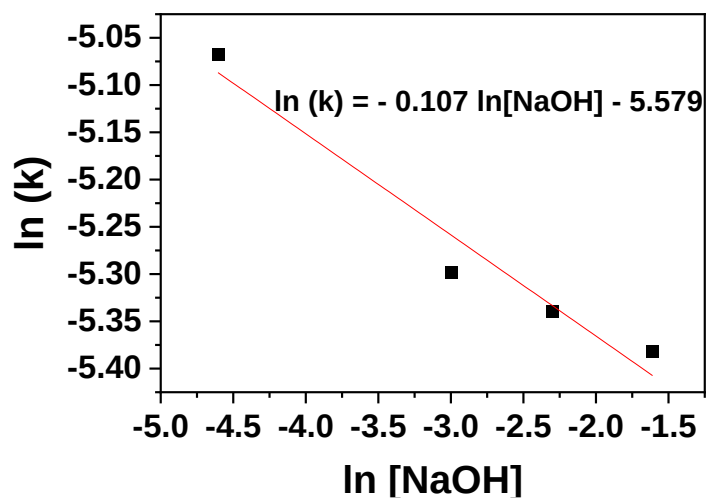


Figure S4. $\ln k$ vs. $\ln [\text{NaOH}]$. Experimental conditions: 50 mg of support, dose of 0.025 mmol Pt (0.0049g), 1.0 mL NaBH₄ (0.500 mol L⁻¹), constant agitation, and temperature of 293.15 K.

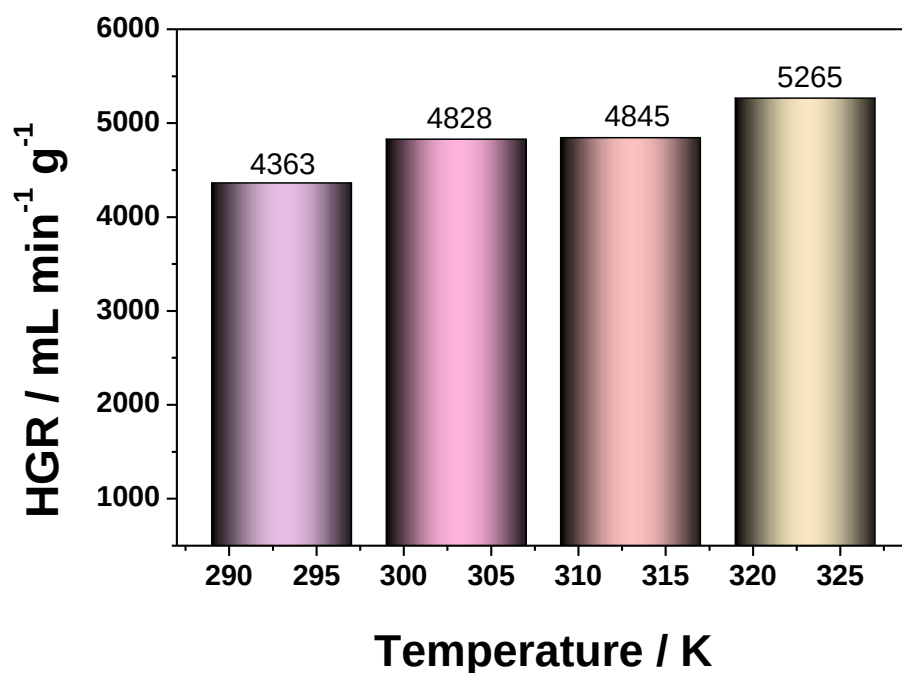


Figure S5. Hydrogen generation rate (HGR) of Pt NPs/MnWO₄. Experimental conditions: 50 mg of support, dose of 0.025 mmol Pt (0.0049g), 1.0 mL NaBH₄ (0.500 mol L⁻¹), constant agitation.

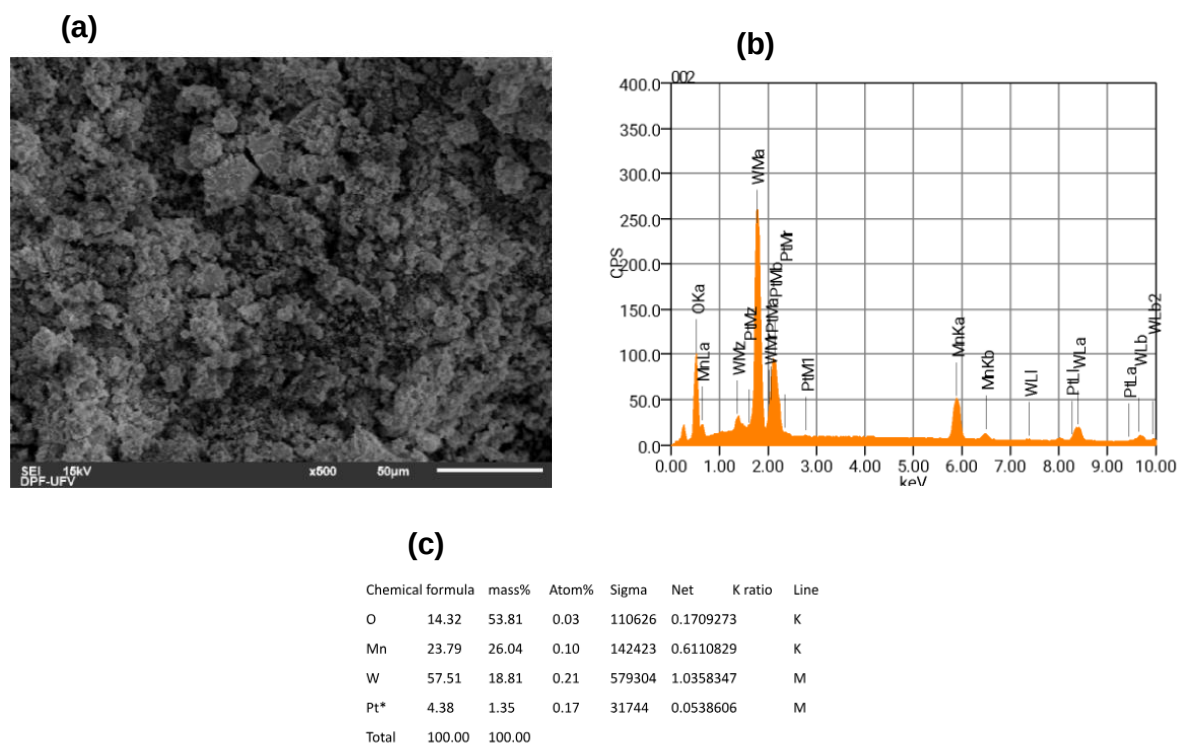


Figure S6. (a) Scanning electron microscopy (SEM), (b) energy-dispersive X-ray spectroscopy (EDS), and (c) composition analysis of Pt NPs/MnWO₄ after multiple use cycles.