

Supplementary Information

Cytotoxicity and Leishmanicidal Activity of Isoniazid-Derived Hydrazones and 2-Pyrazineformamide Thiosemicarbazones

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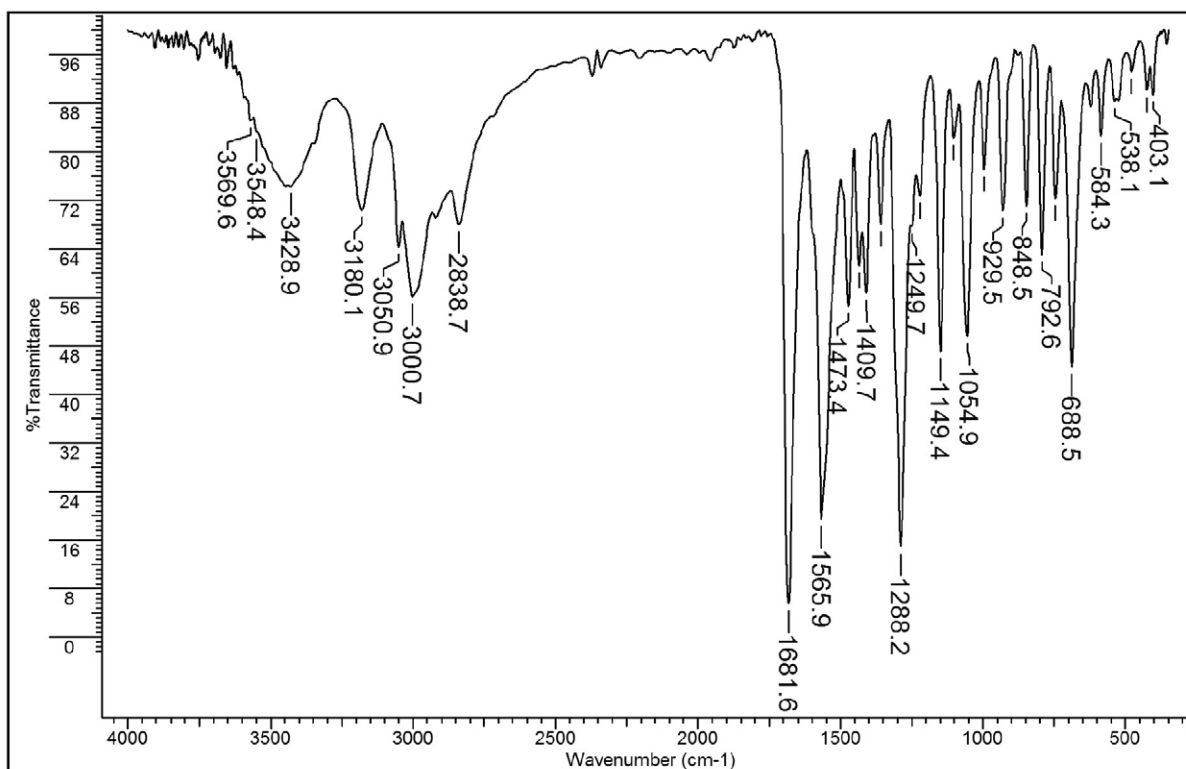


Figure S1. FTIR (KBr) spectrum of compound 1.

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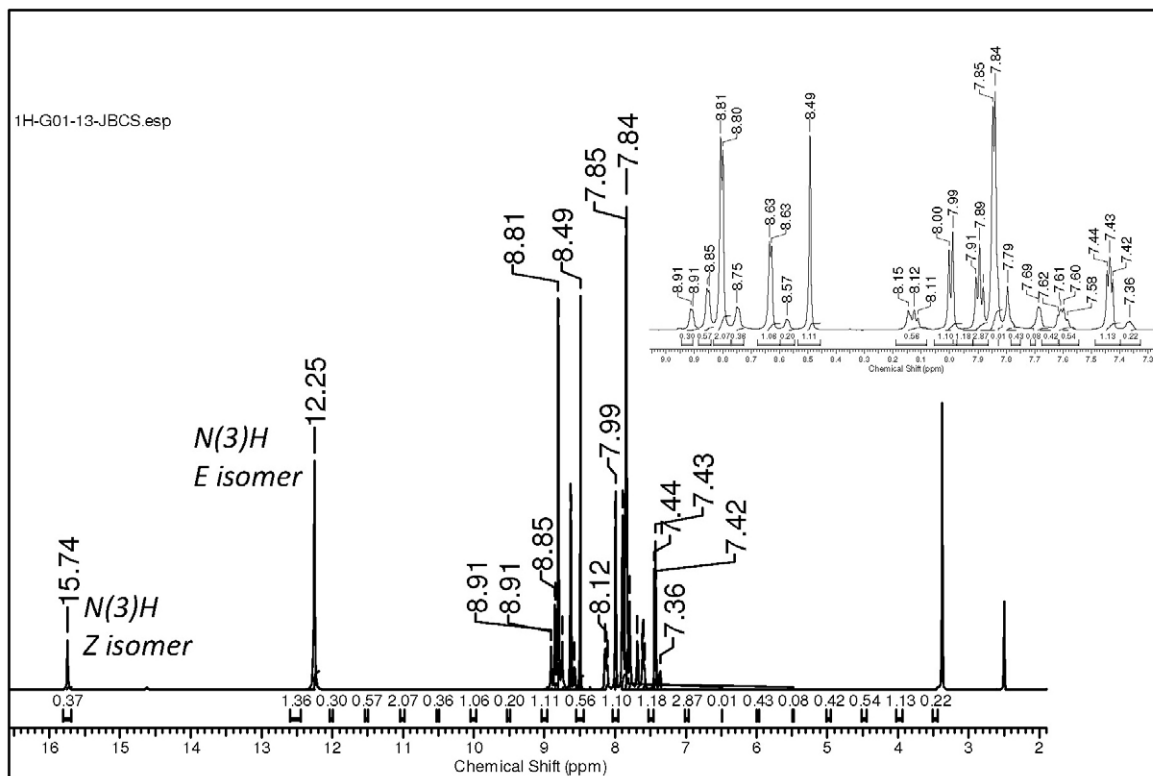


Figure S2. ^1H NMR spectrum (600 MHz, $\text{DMSO}-d_6$) of compound 1.

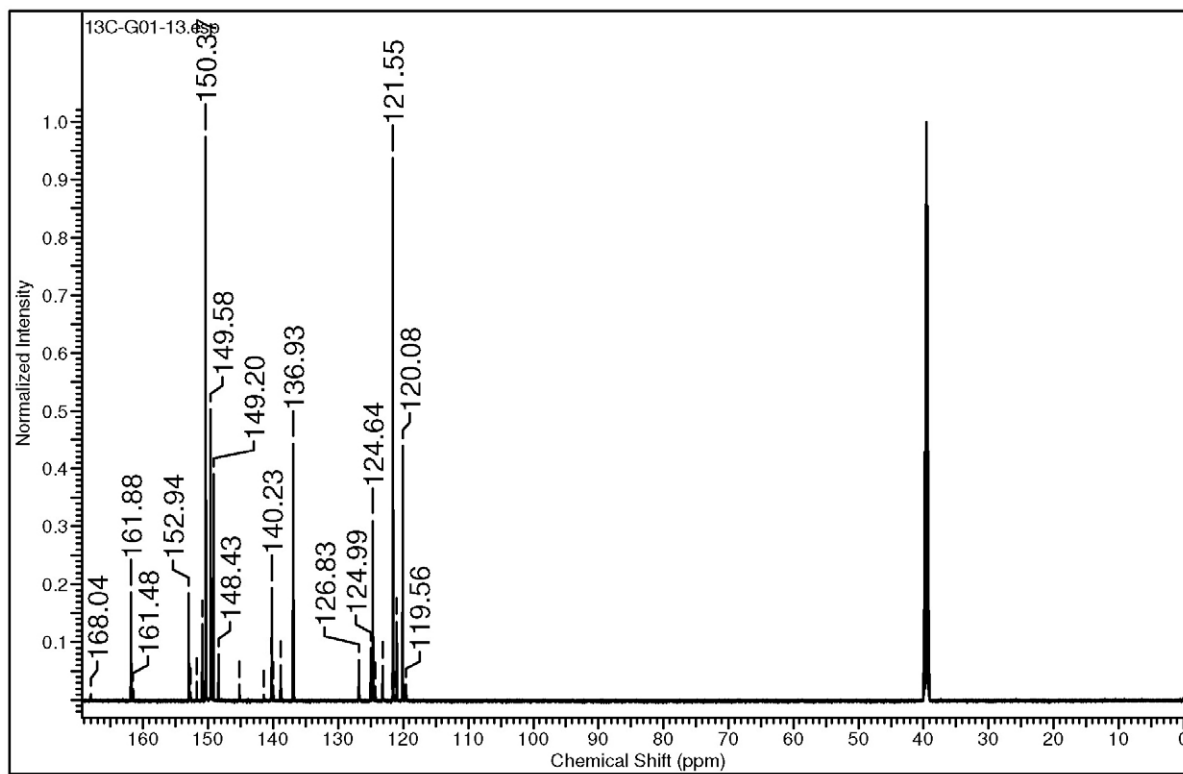


Figure S3. ^{13}C NMR spectrum (150 MHz, $\text{DMSO}-d_6$) of compound 1.

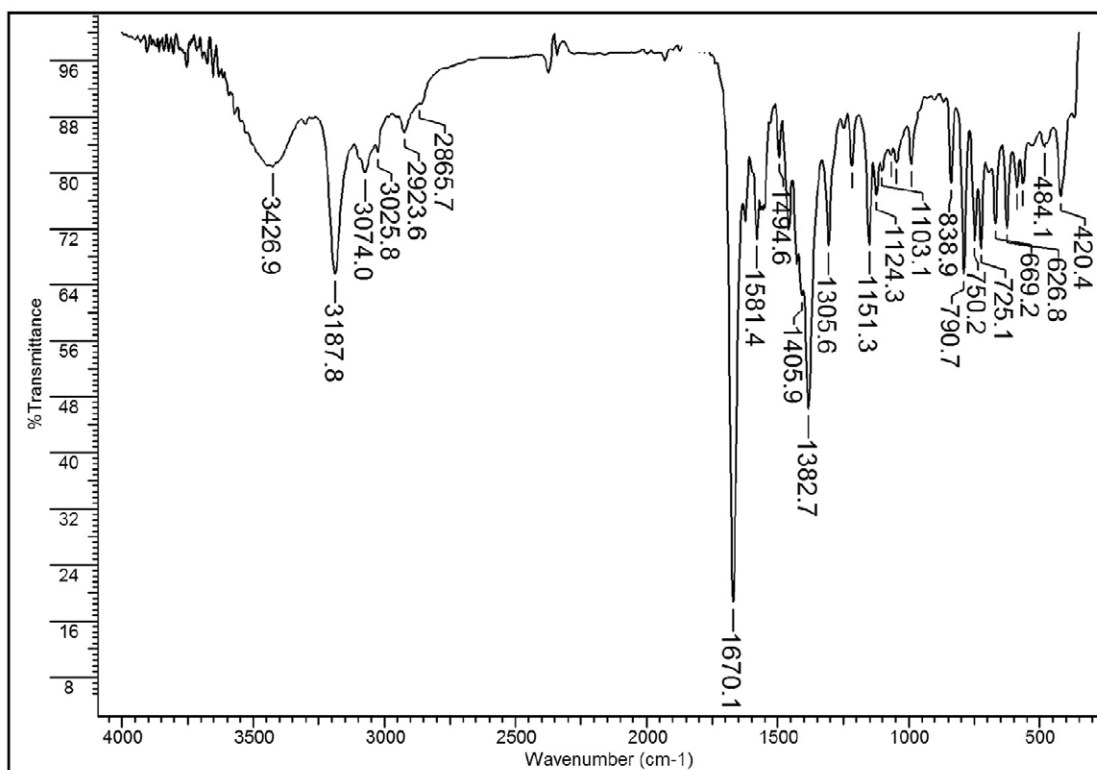


Figure S4. FTIR (KBr) spectrum of compound **2**.

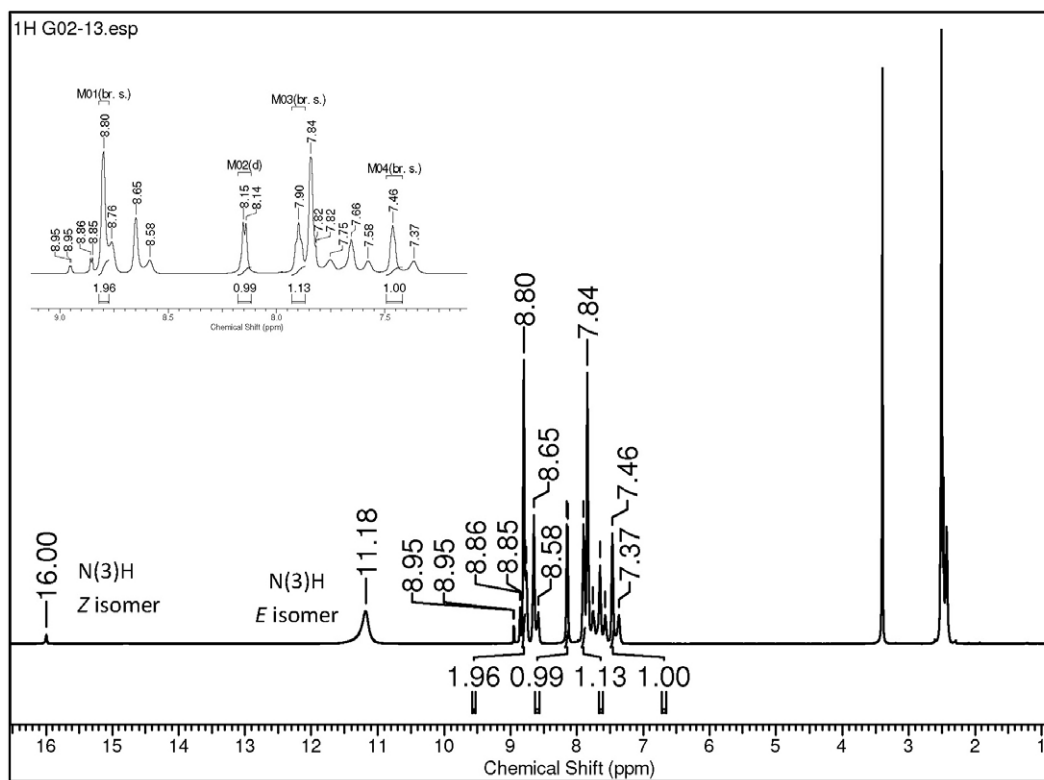


Figure S5. ^1H NMR spectrum (600 MHz, $\text{DMSO}-d_6$) of compound **2**.

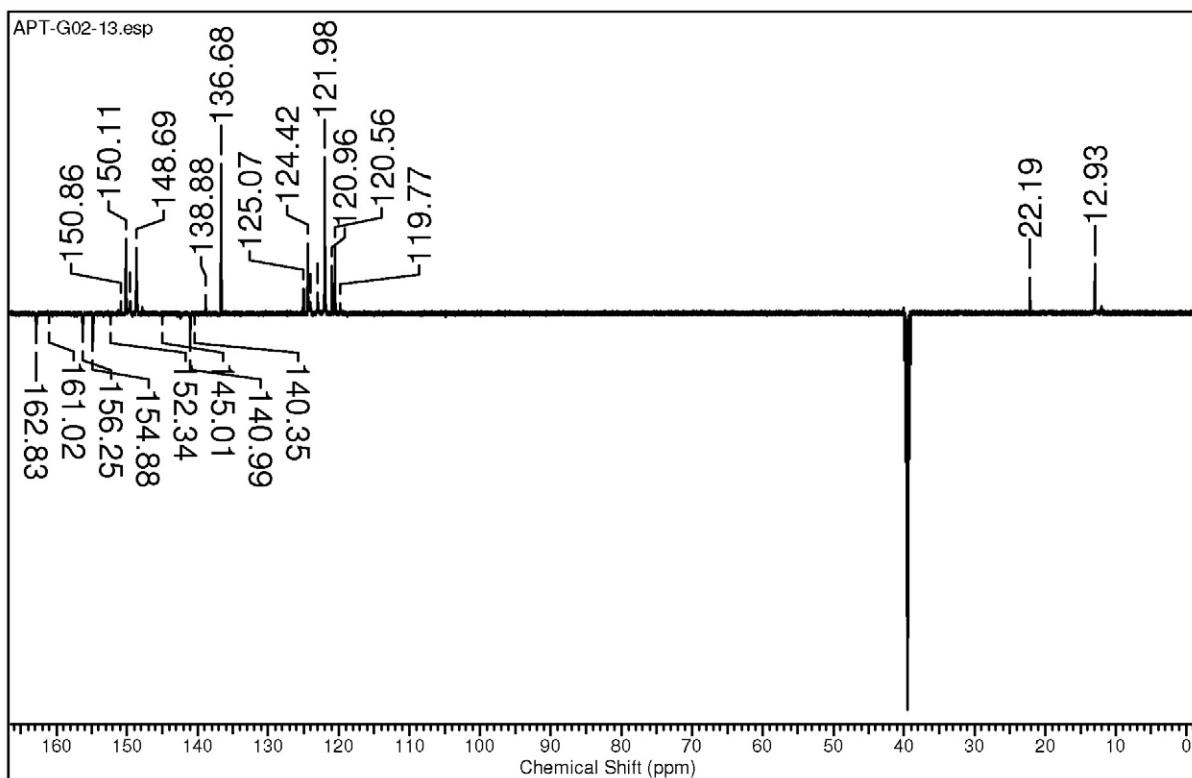


Figure S6. ^{13}C NMR spectrum (150 MHz, $\text{DMSO}-d_6$) of compound 2.

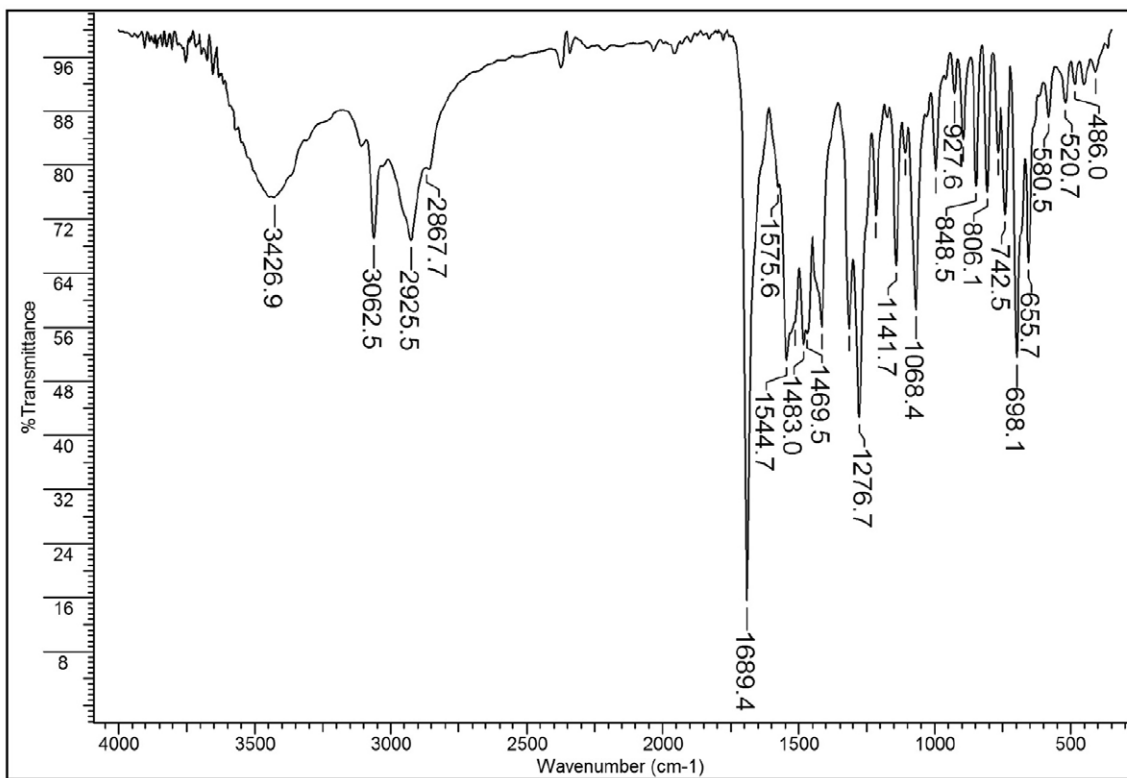


Figure S7. FTIR (KBr) spectrum of compound 3.

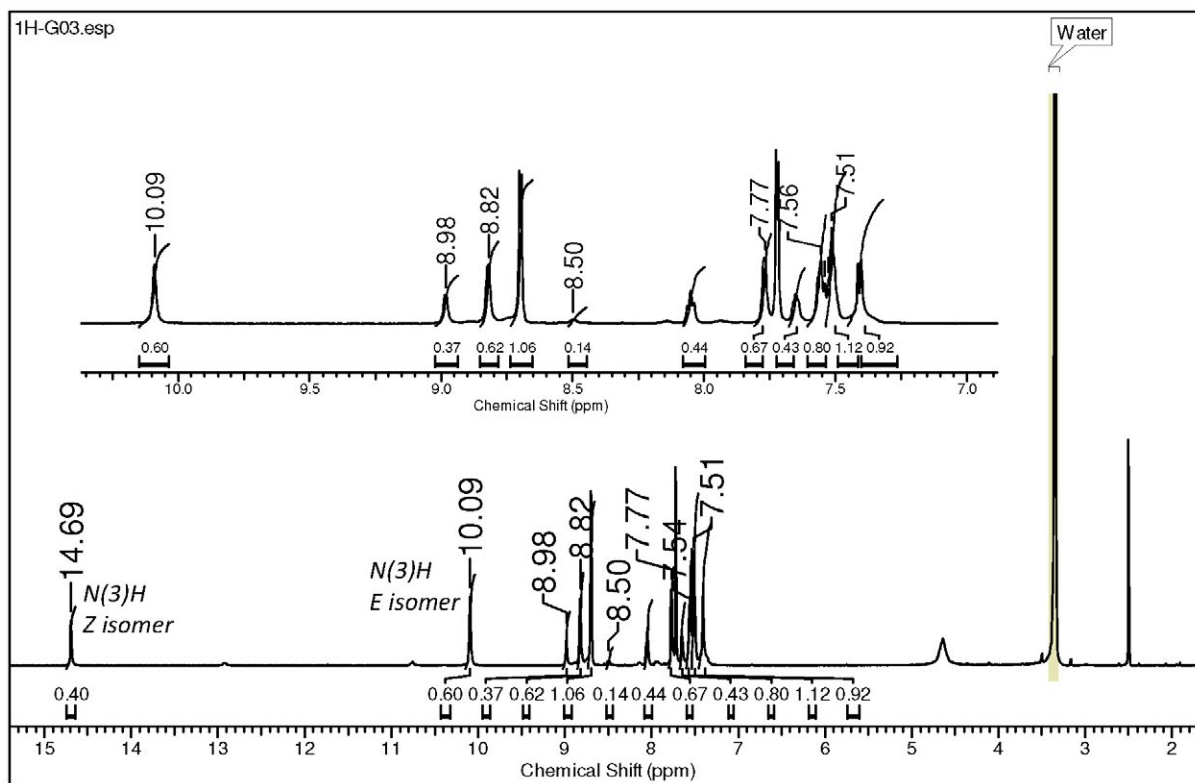


Figure S8. ^1H NMR spectrum (600 MHz, $\text{DMSO}-d_6$) of compound 3.

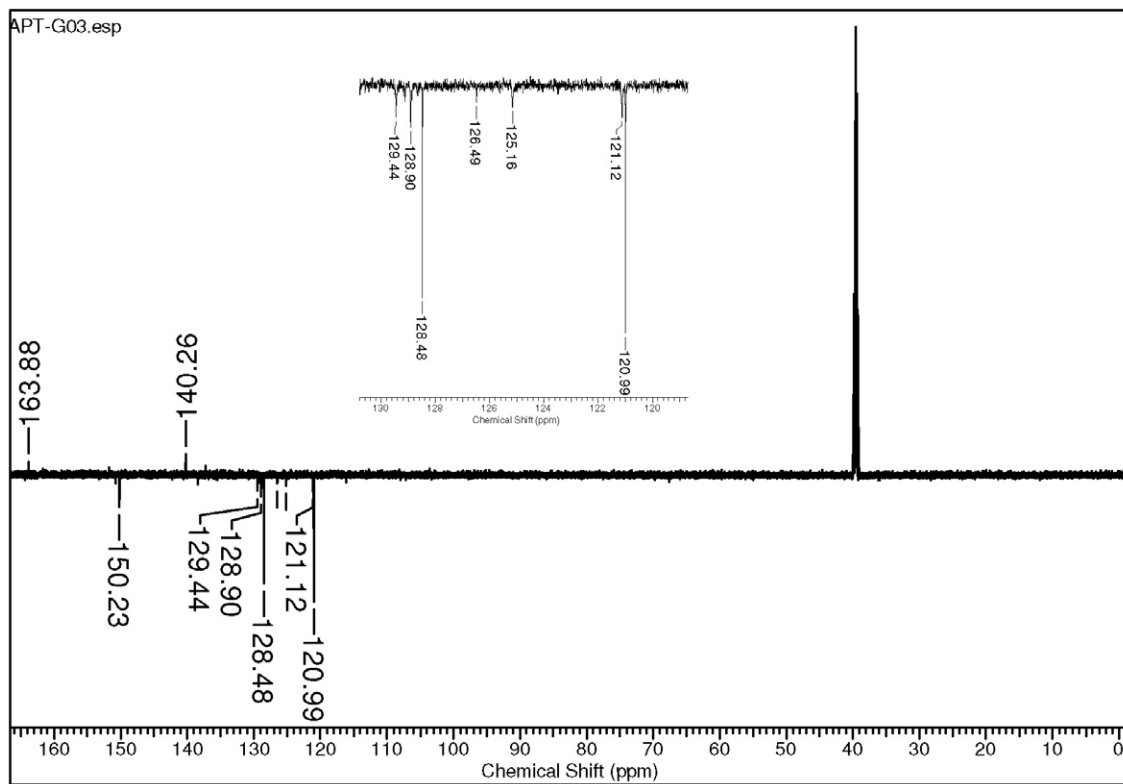


Figure S9. ^{13}C NMR spectrum (150 MHz, $\text{DMSO}-d_6$) of compound 3.

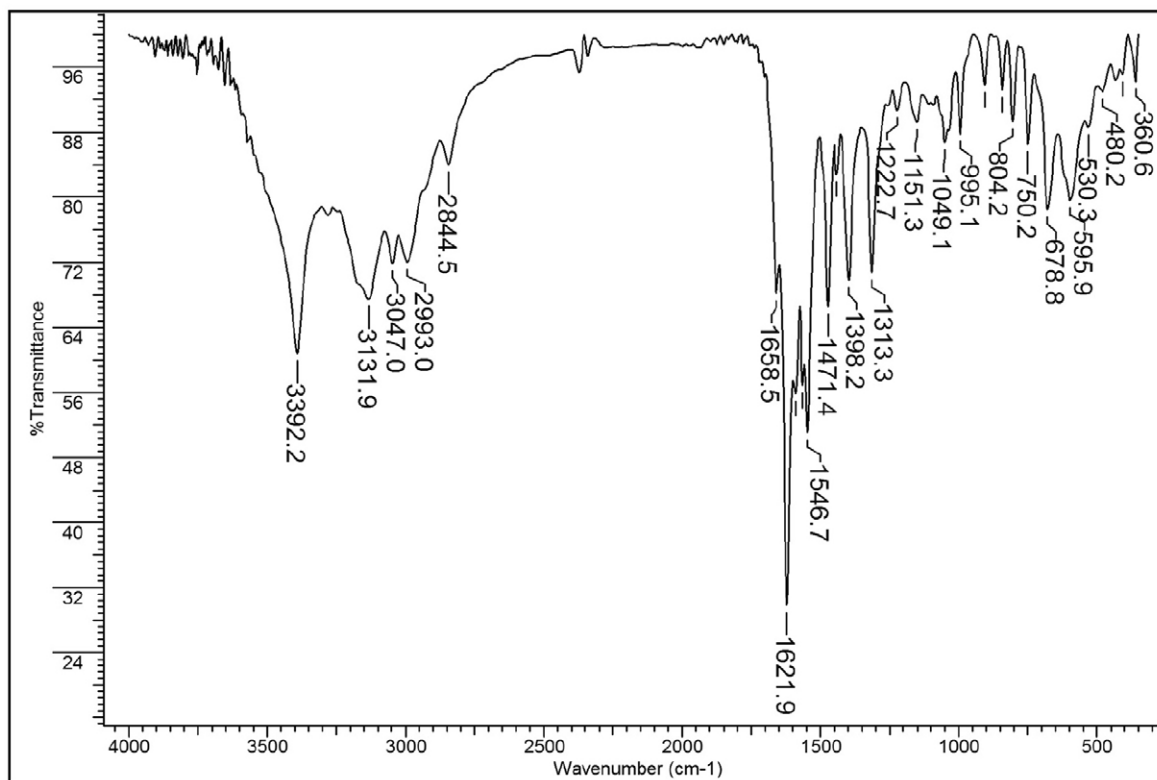


Figure S10. FTIR (KBr) spectrum of compound 4.

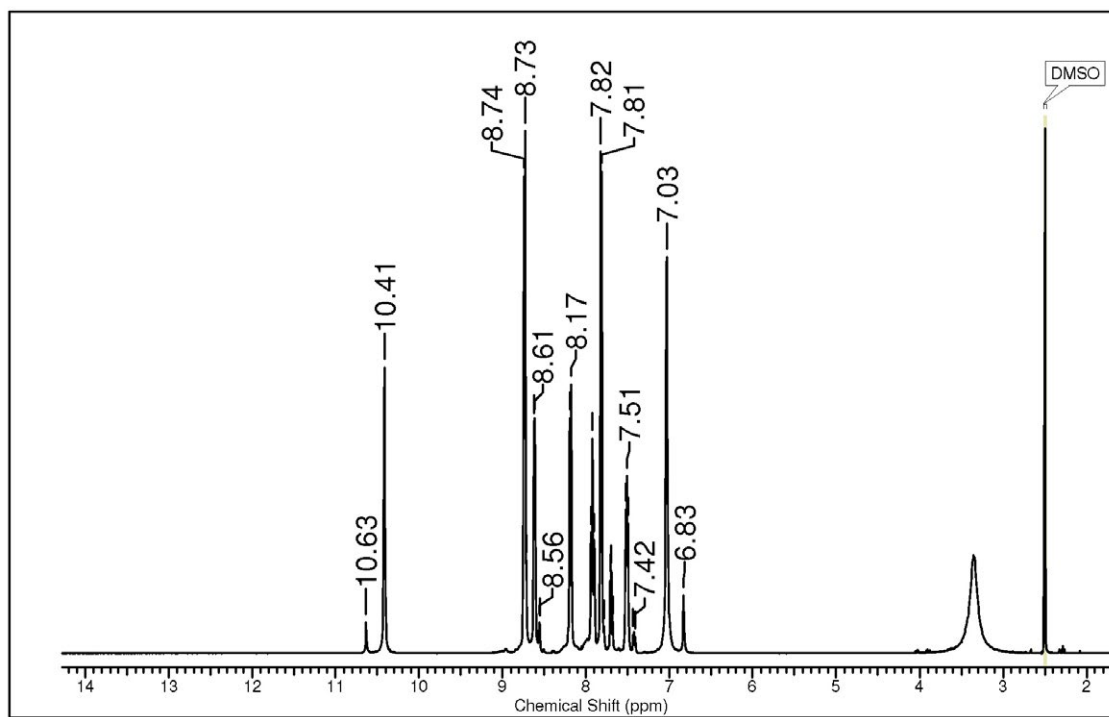


Figure S11. ¹H NMR spectrum (400 MHz, DMSO-*d*₆) of compound 4.

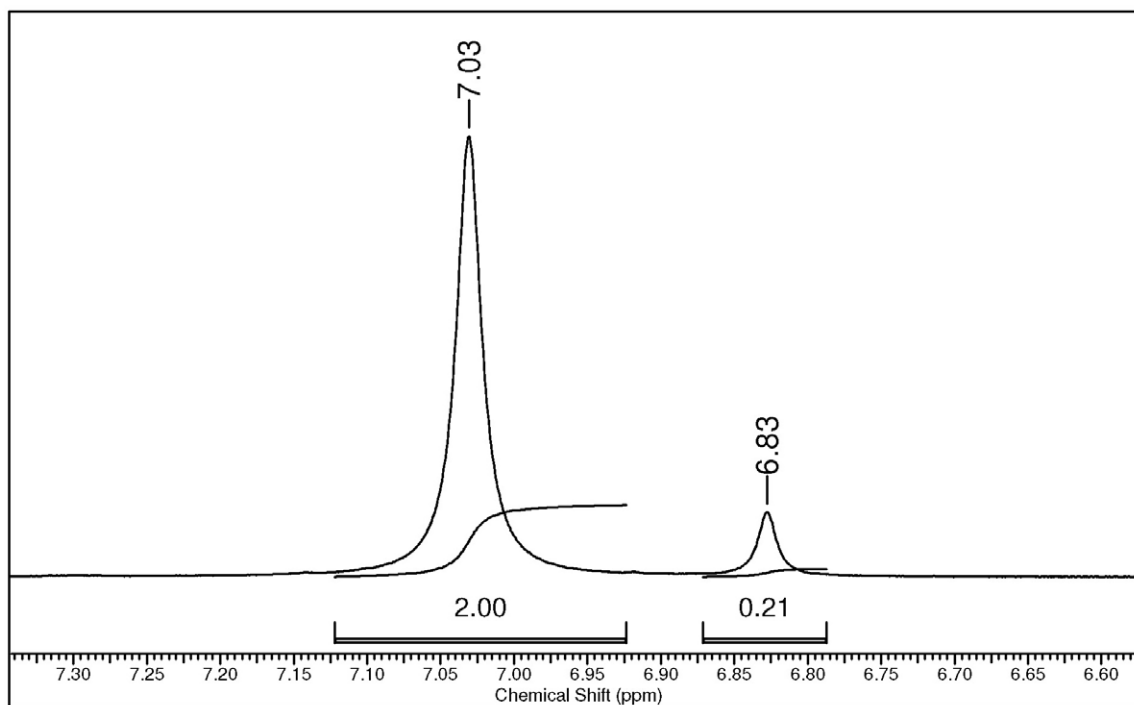


Figure S12. ¹H NMR spectrum (400 MHz, DMSO-*d*₆) of compound **4** (6.60-7.30 ppm region).

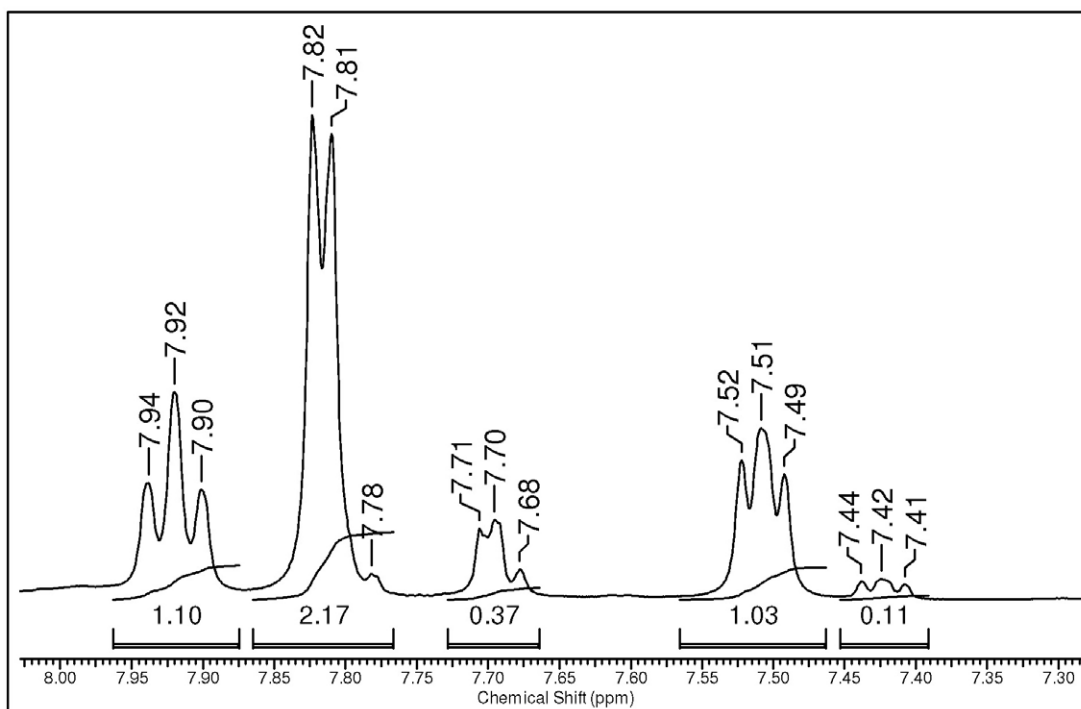


Figure S13. ¹H NMR spectrum (400 MHz, DMSO-*d*₆) of compound **4** (7.30-8.00 ppm region).

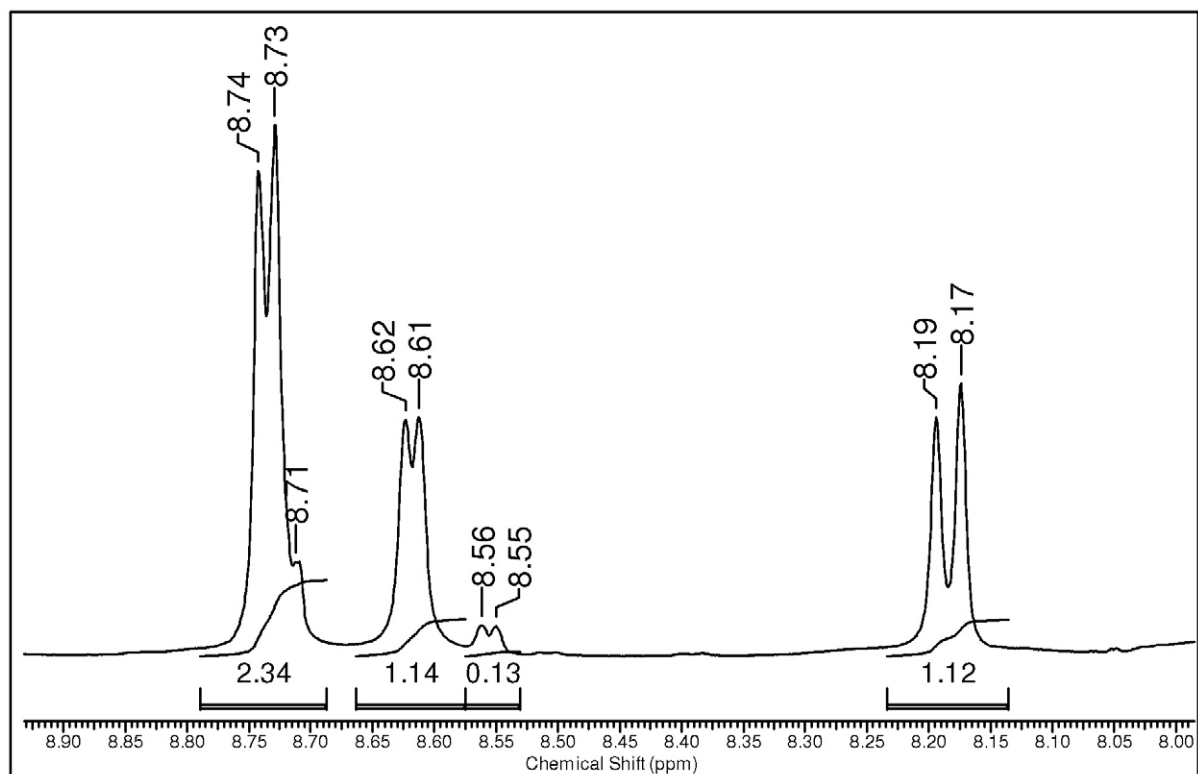


Figure S14. ¹H NMR spectrum (400 MHz, DMSO-*d*₆) of compound 4 (8.00-8.90 ppm region).

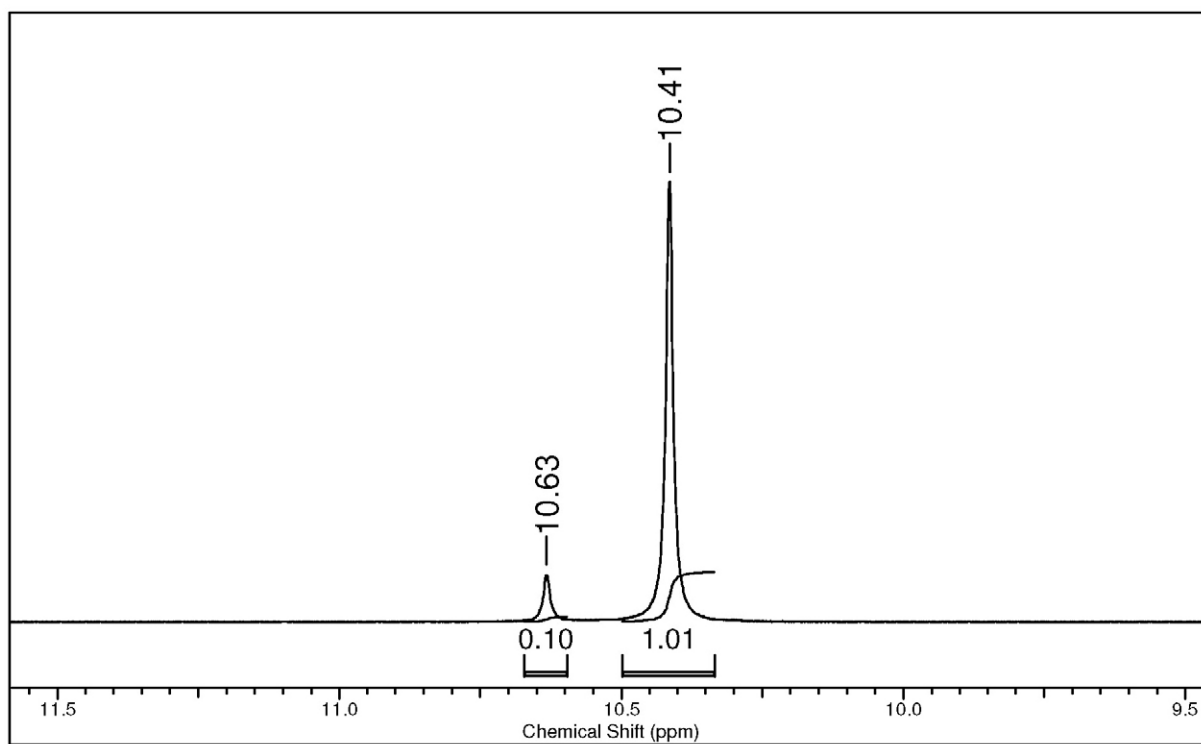


Figure S15. ¹H NMR spectrum (400 MHz, DMSO-*d*₆) of compound 4 (9.5-11.5 ppm region).

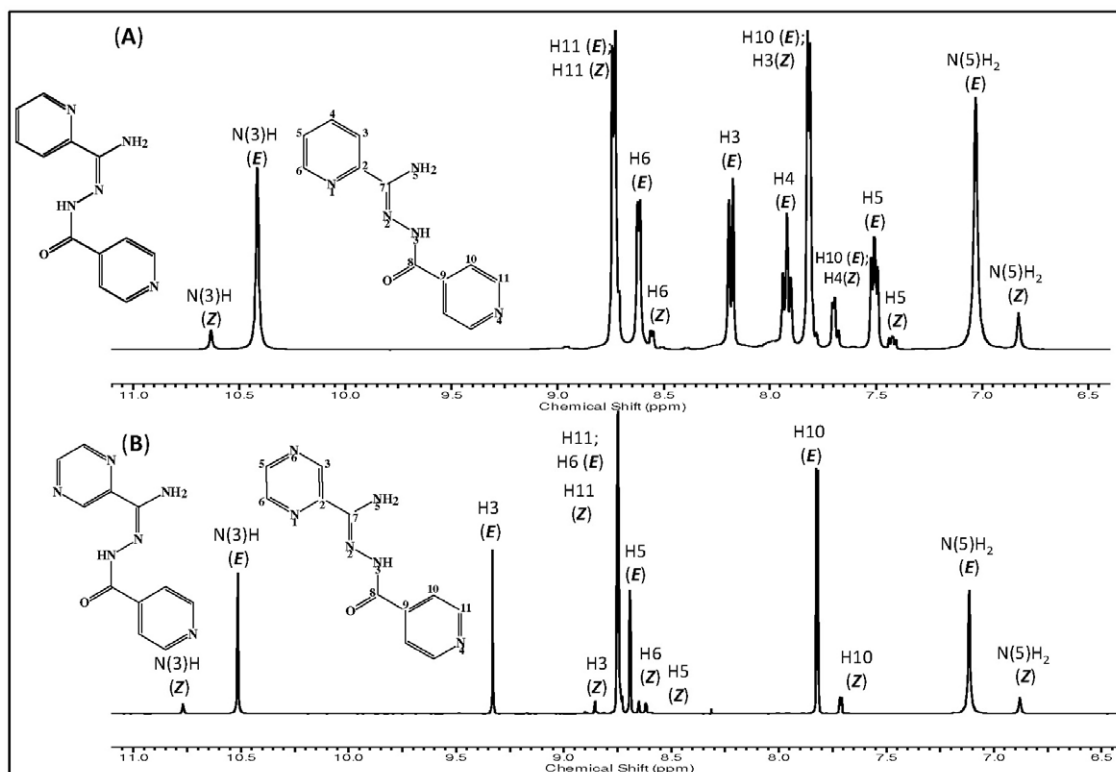


Figure S16. ^1H NMR spectra of **4** (A, 400 MHz) and **5** (B, 600 MHz) recorded in $\text{DMSO-}d_6$ (6.5–11.0 ppm region). In both cases, the data suggest the compounds exist as a mixture of *E* and *Z* isomers in solution. These isomers had previously been reported for compound **5**.¹

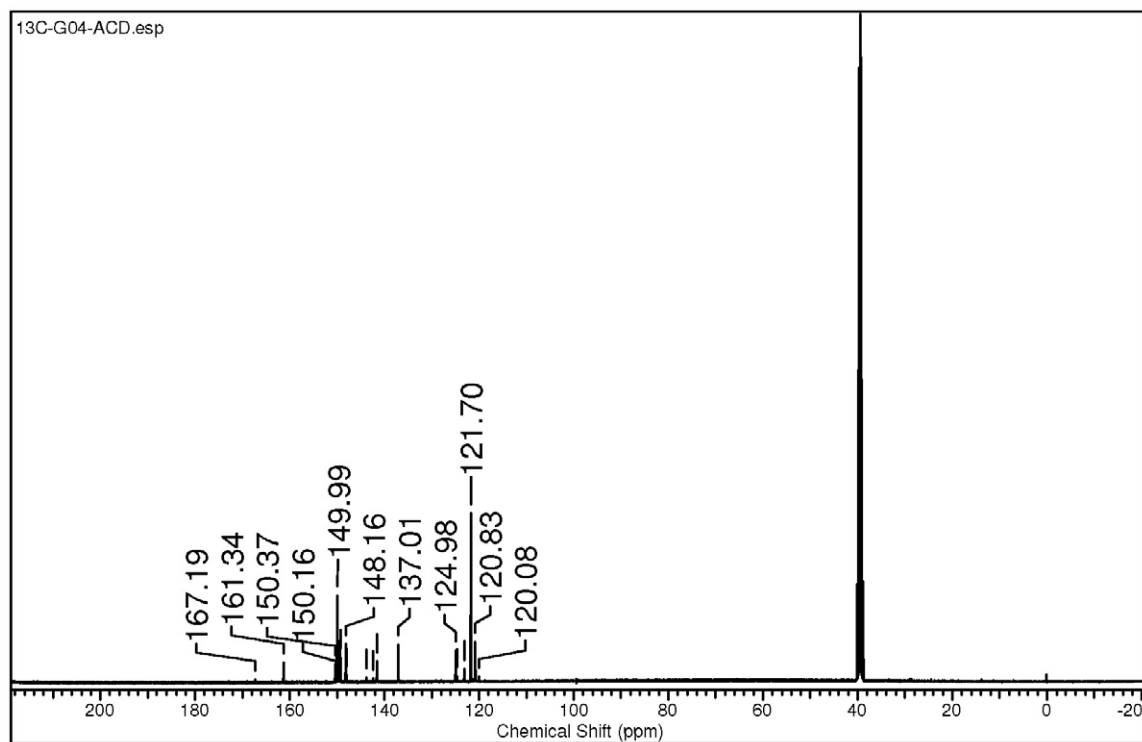


Figure S17. ^{13}C NMR spectrum (100 MHz, $\text{DMSO-}d_6$) of compound **4**.

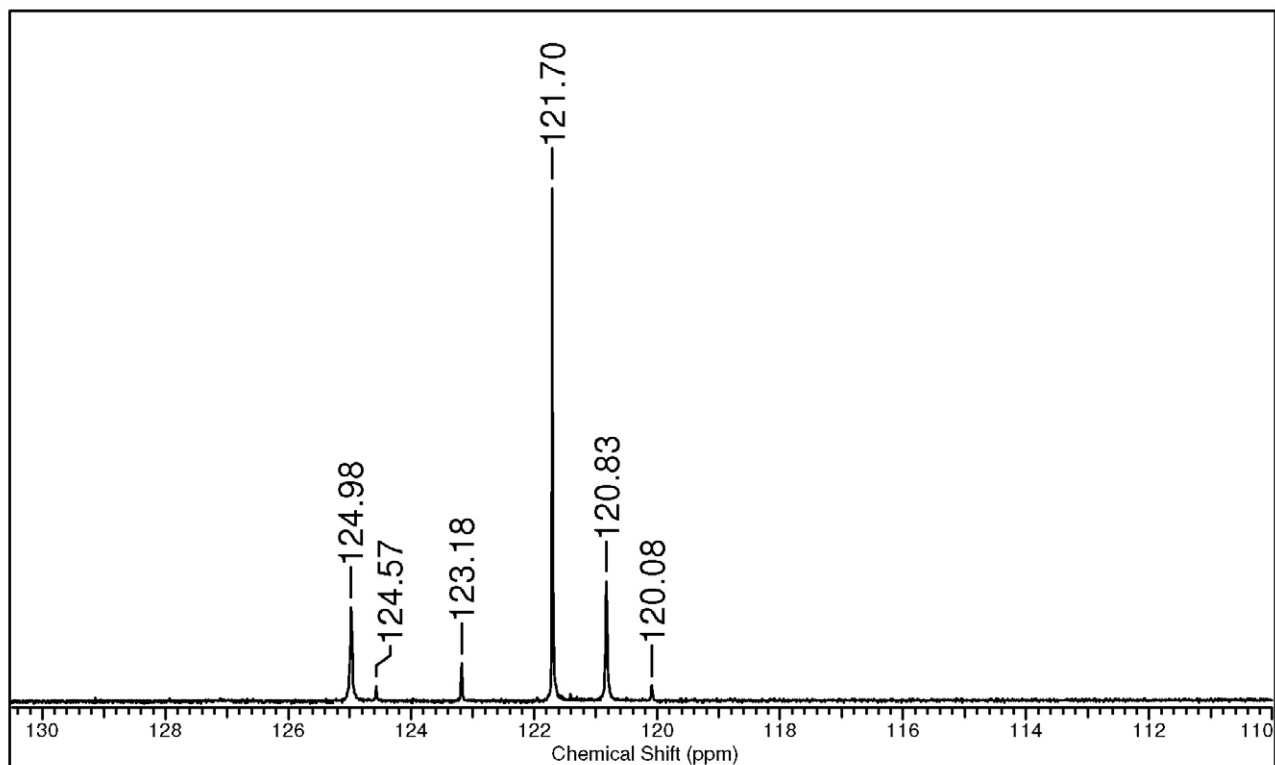


Figure S18. ^{13}C NMR spectrum (100 MHz, $\text{DMSO}-d_6$) of compound 4 (110-130 ppm region).

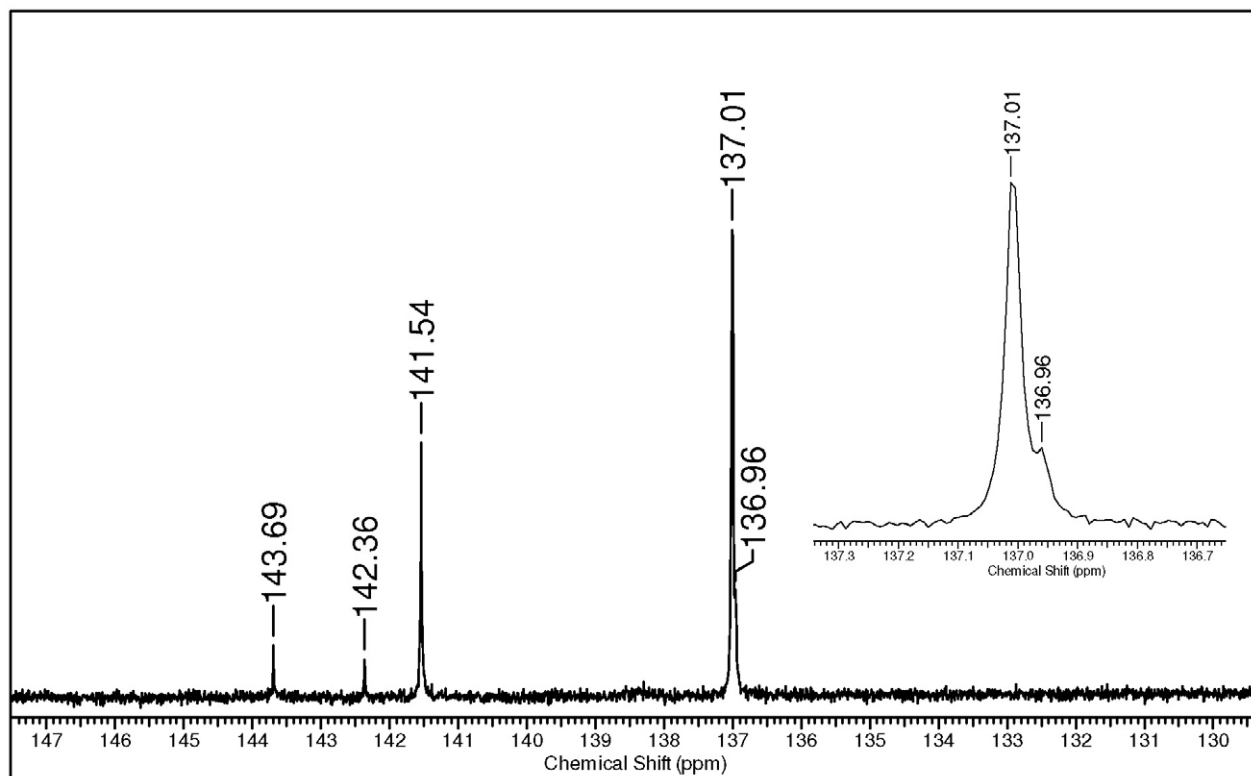


Figure S19. ^{13}C NMR spectrum (100 MHz, $\text{DMSO}-d_6$) of compound 4 (130-147 ppm region).

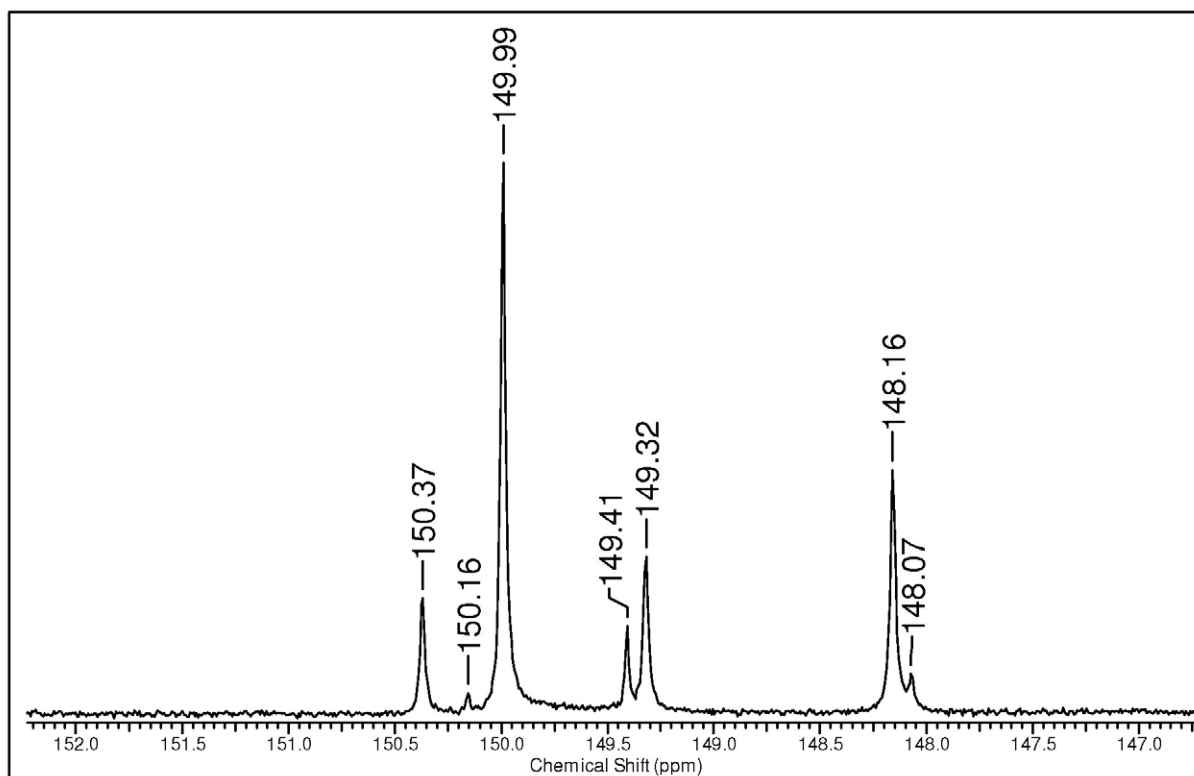


Figure S20. ¹³C NMR spectrum (100 MHz, DMSO-*d*₆) of compound 4 (147-152 ppm region).

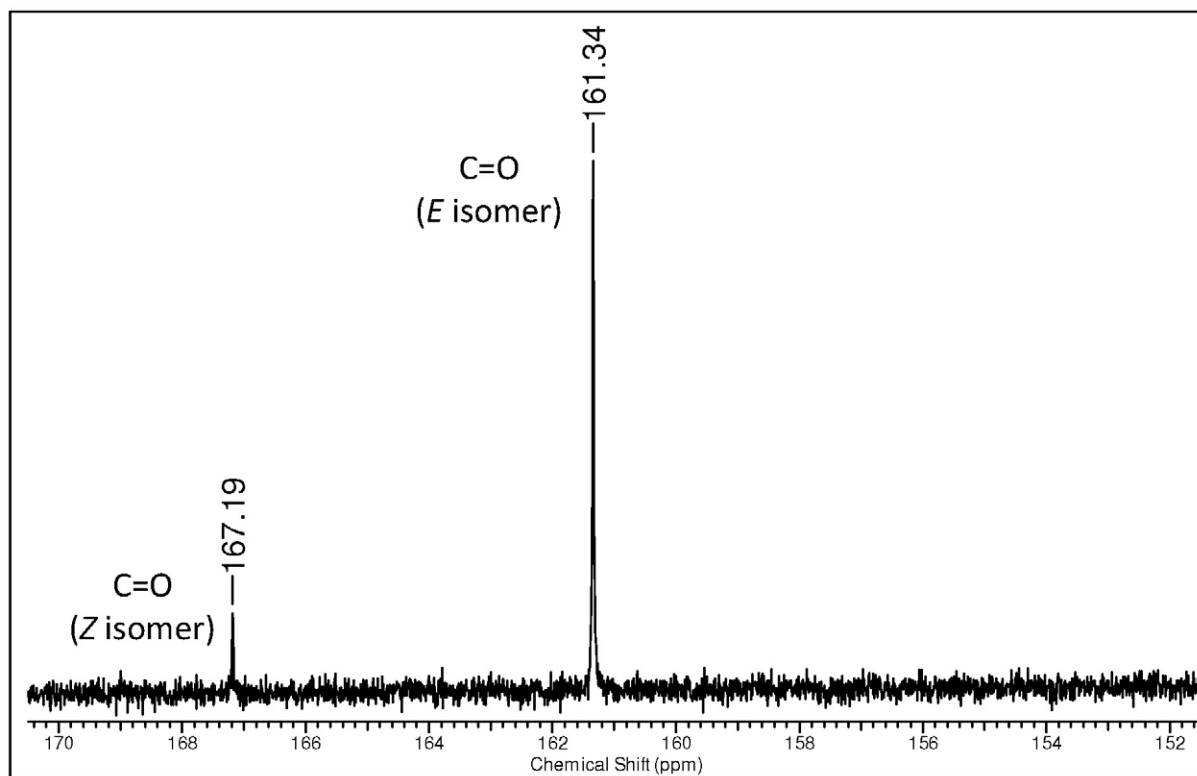


Figure S21. ¹³C NMR spectrum (100 MHz, DMSO-*d*₆) of compound 4 (152-170 ppm region).

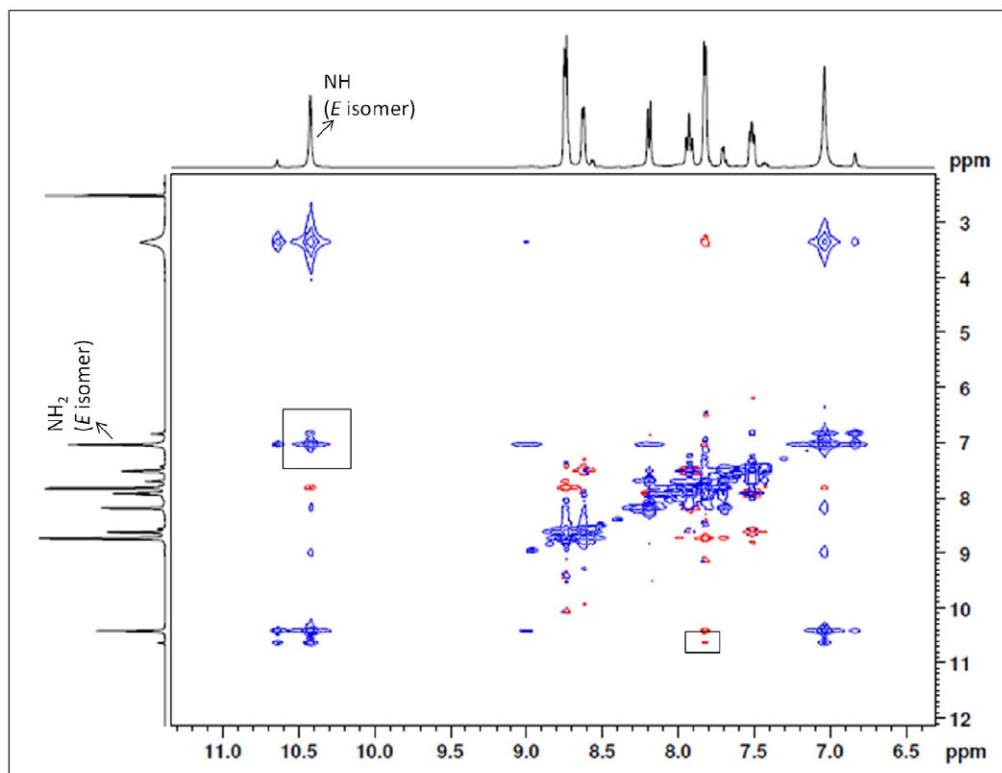


Figure S22. NOESY spectrum (400 MHz, DMSO- d_6) of compound 4.

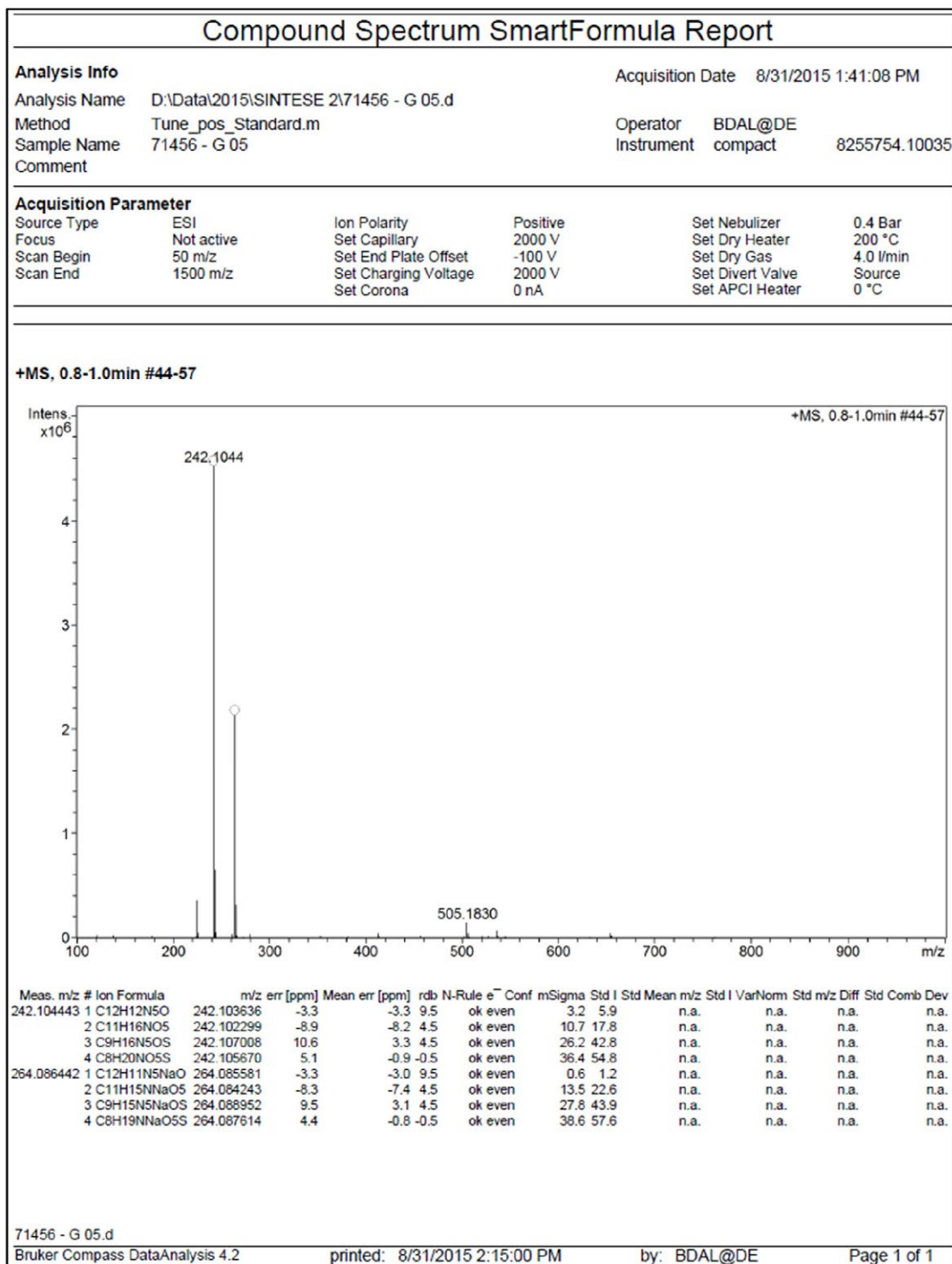


Figure S23. Mass spectrum (ESI-MS) of compound 4.

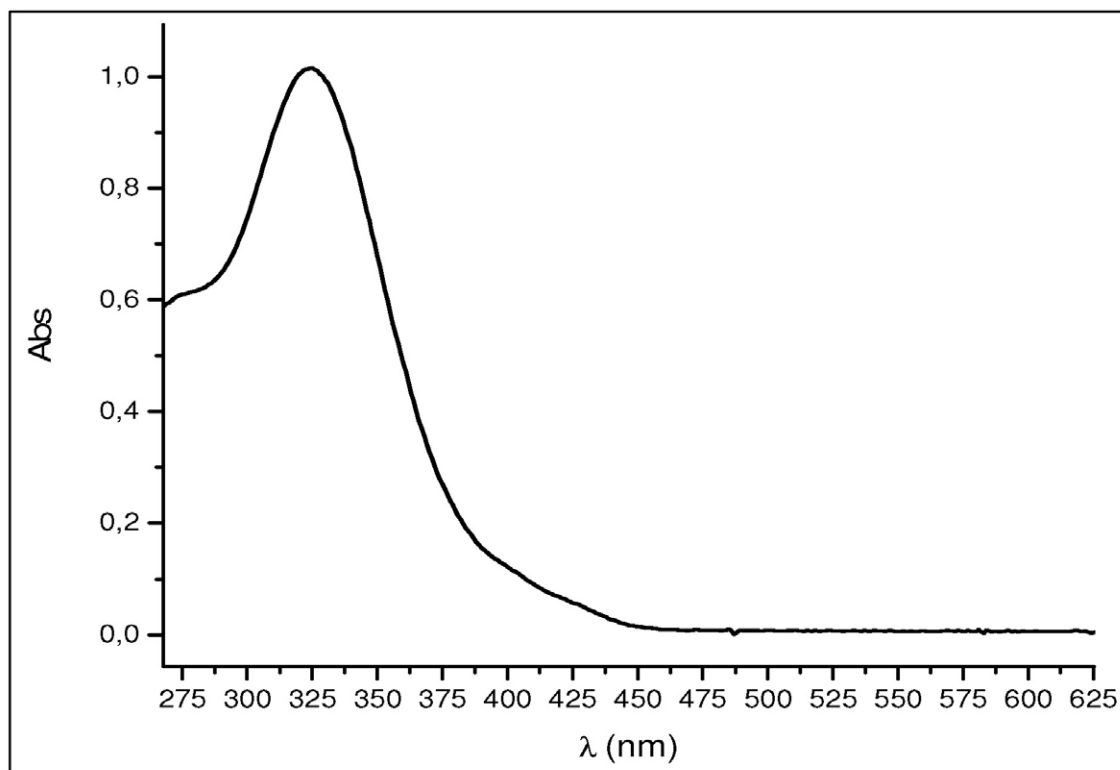


Figure S24. UV-Vis spectrum (DMSO) of compound 4.

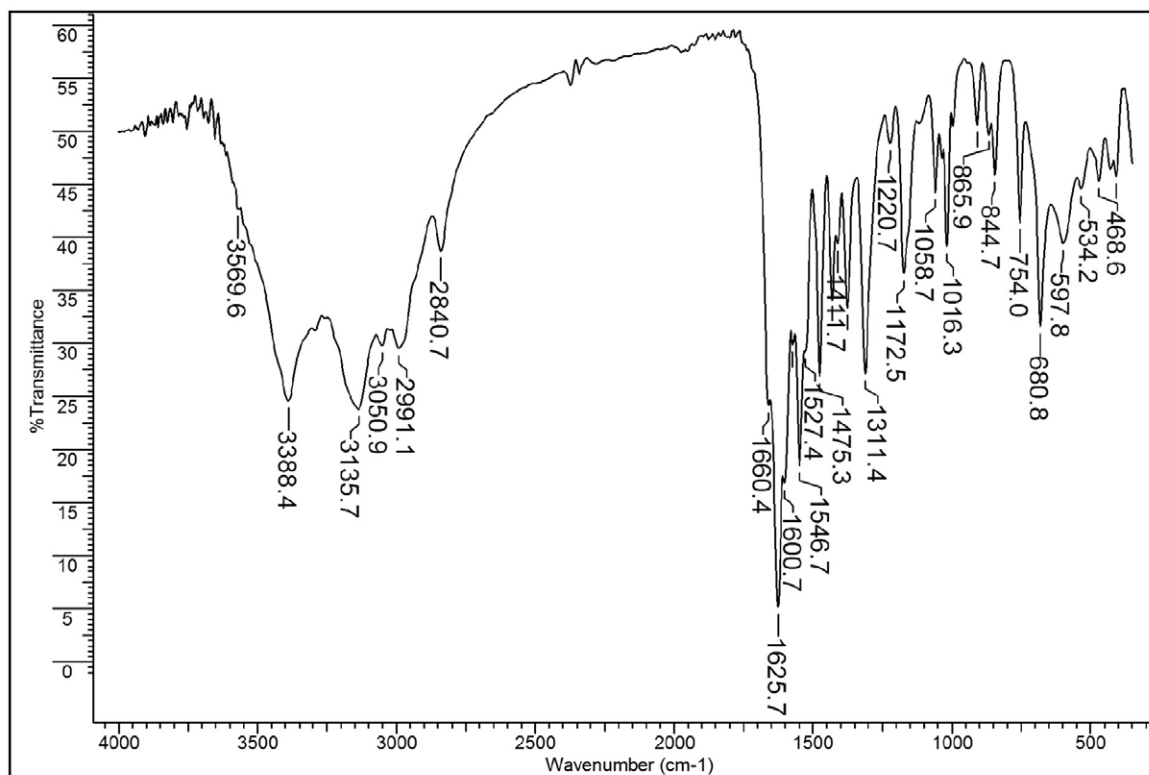


Figure S25. FTIR (KBr) spectrum of compound 5.

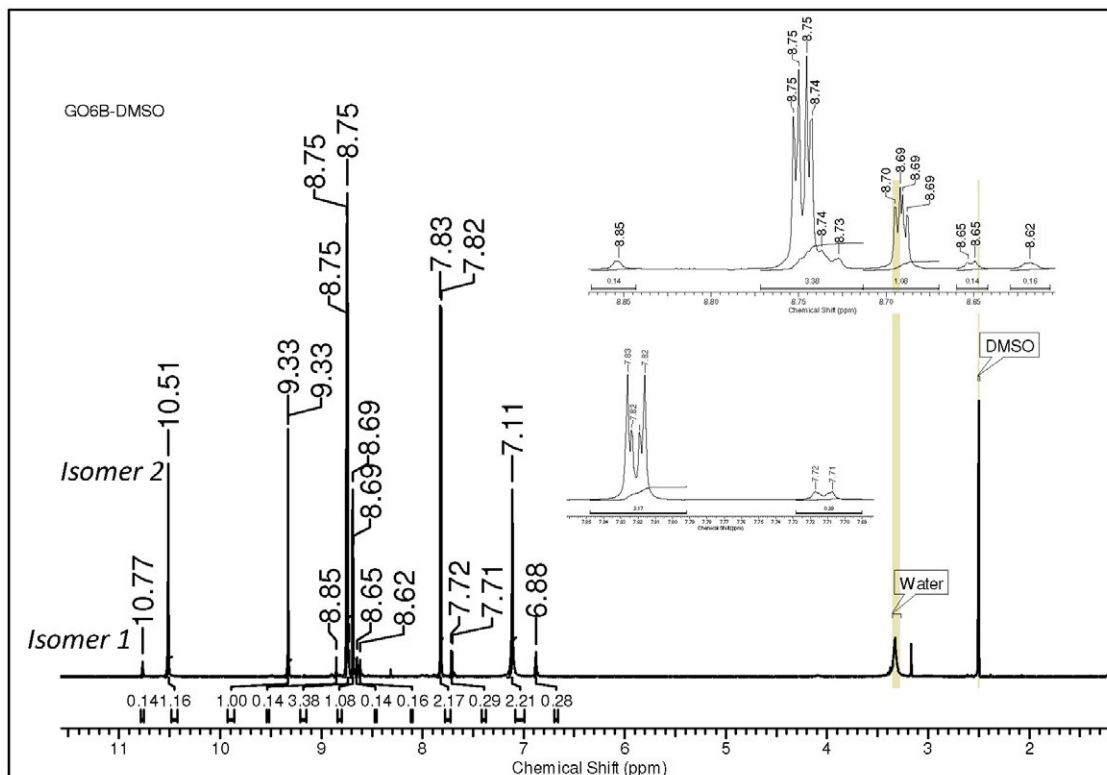


Figure S26. ^1H NMR spectrum (600 MHz, $\text{DMSO-}d_6$) of compound 5.

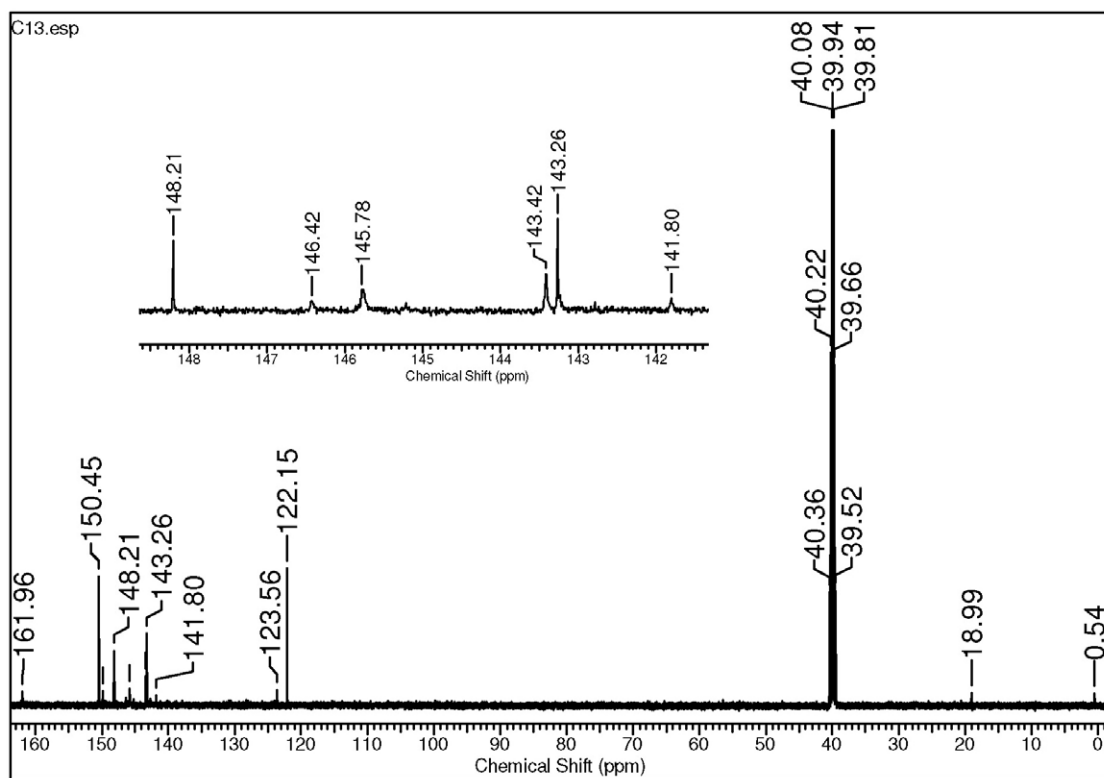


Figure S27. ^{13}C NMR spectrum (150 MHz, $\text{DMSO-}d_6$) of compound 5.

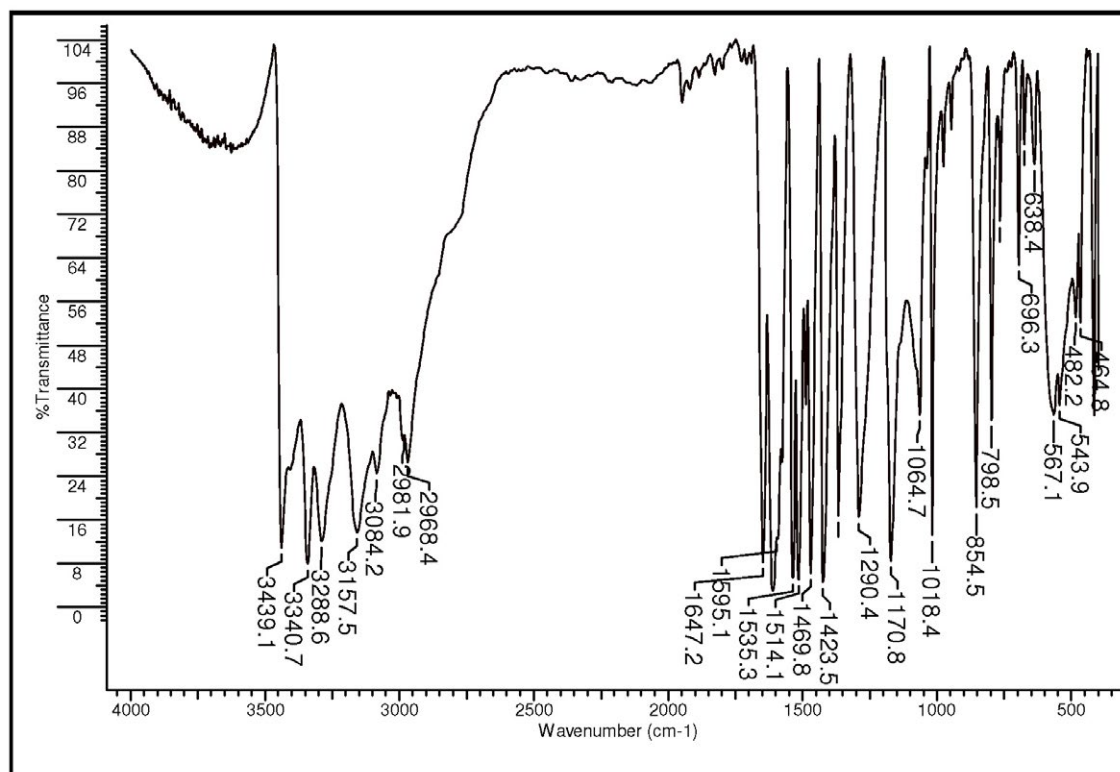


Figure S28. FTIR (KBr) spectrum of compound 6.

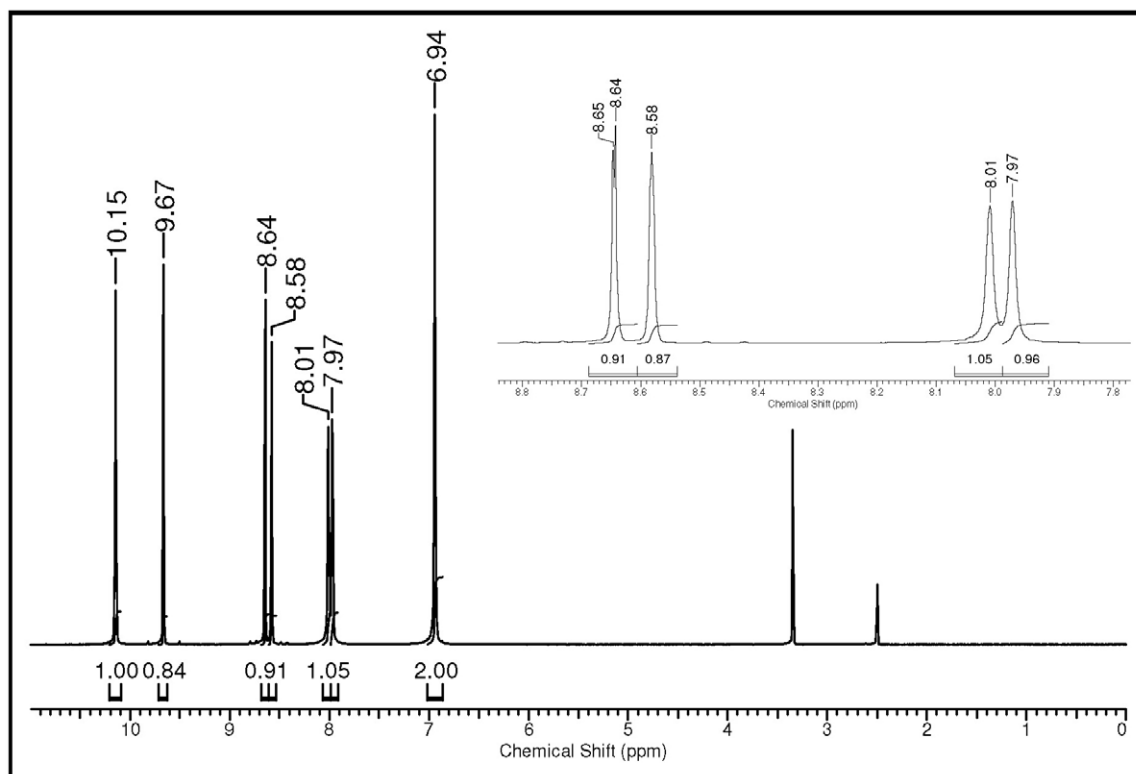


Figure S29. ¹H NMR spectrum (600 MHz, DMSO-*d*₆) of compound 6.

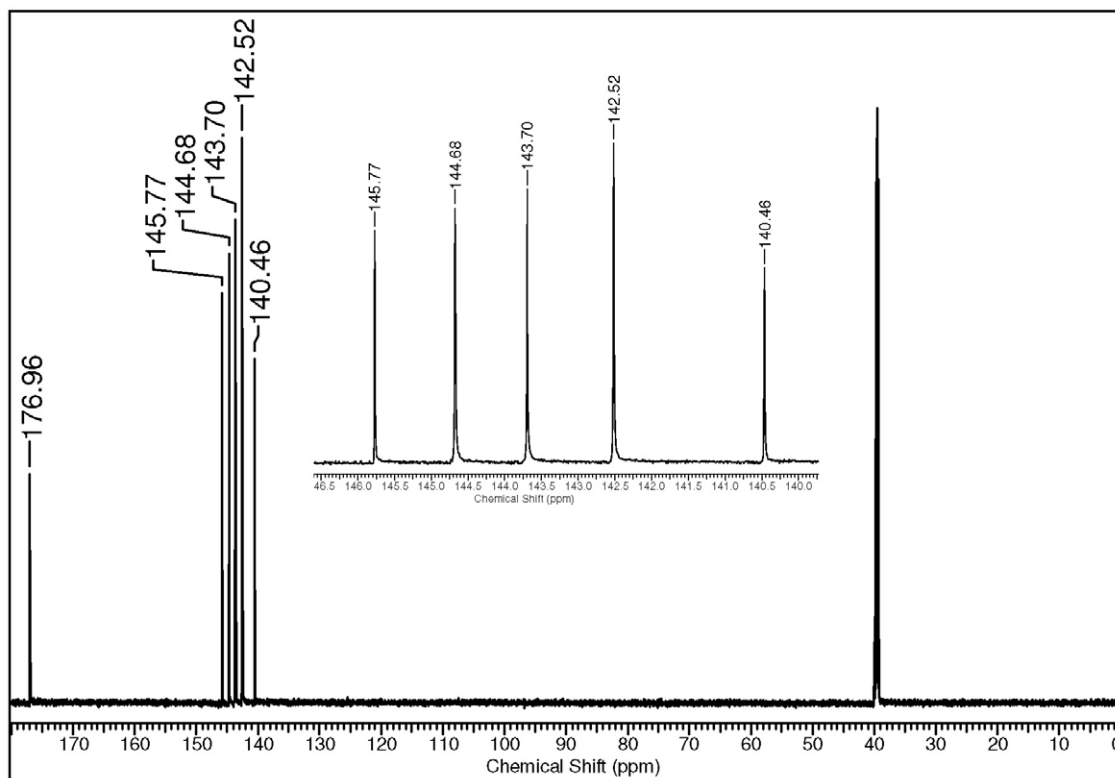


Figure S30. ^{13}C NMR spectrum (150 MHz, $\text{DMSO}-d_6$) of compound 6.

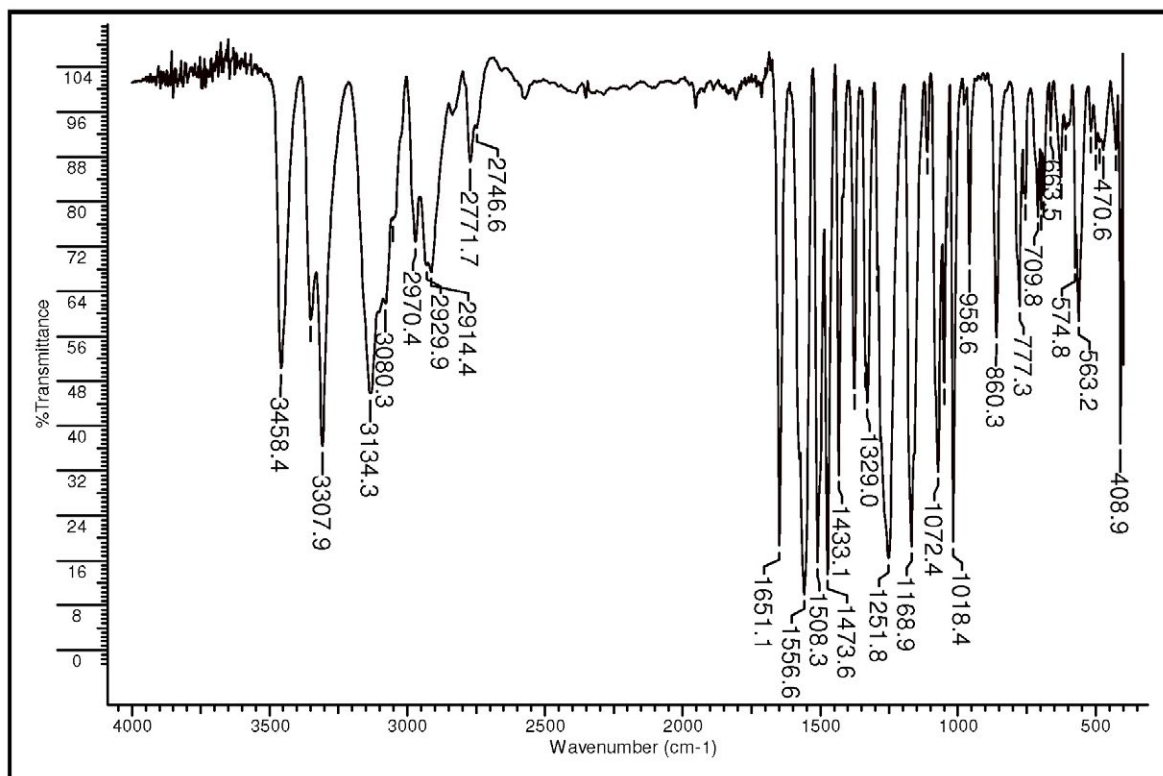


Figure S31. FTIR (KBr) spectrum of compound 7.

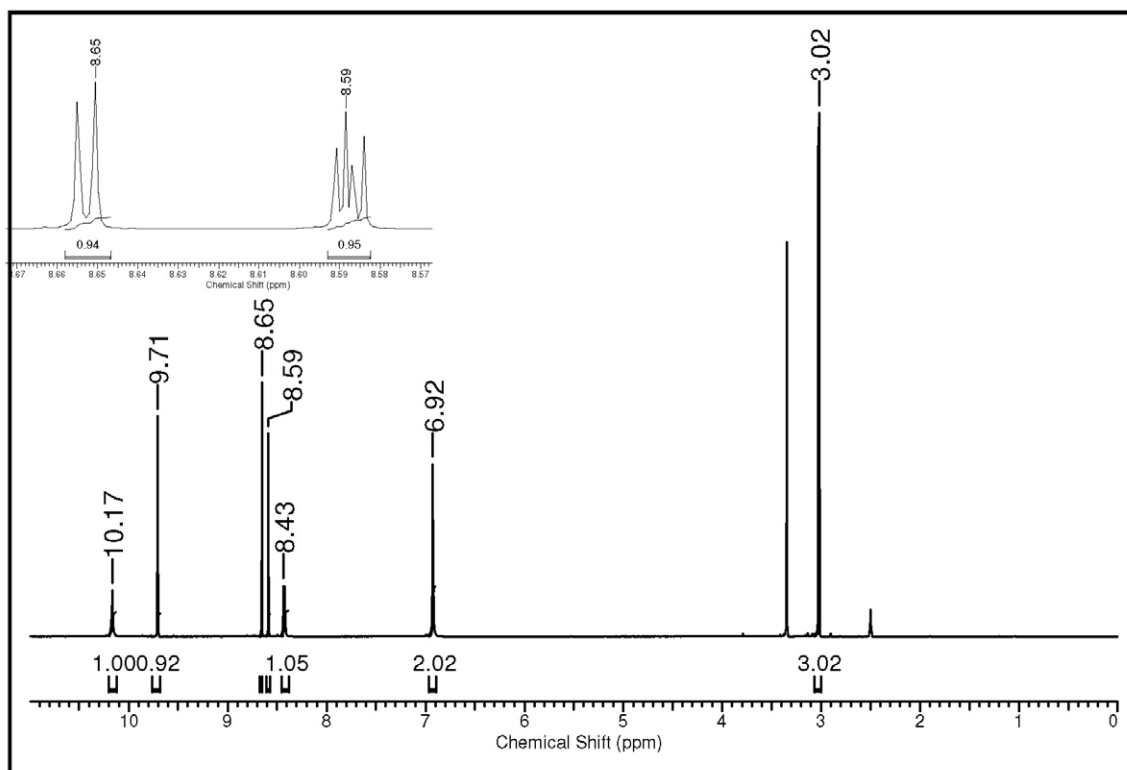


Figure S32. ^1H NMR spectrum (600 MHz, $\text{DMSO}-d_6$) of compound 7.

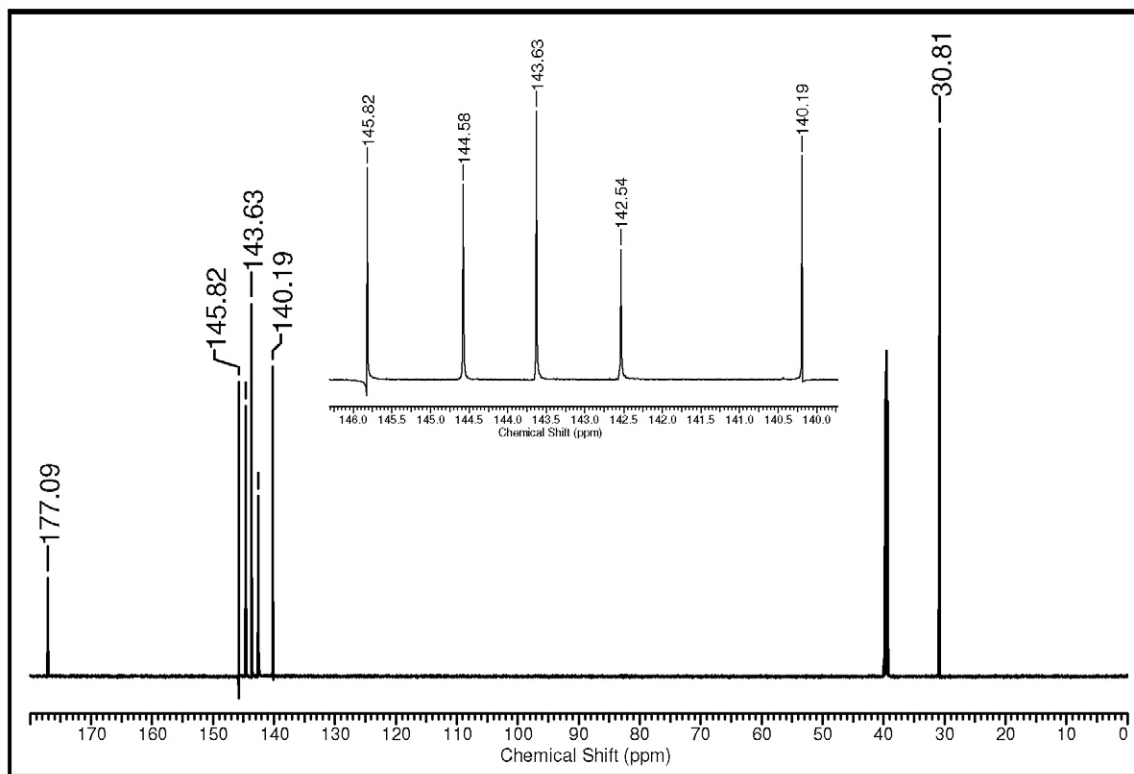


Figure S33. ^{13}C NMR spectrum (150 MHz, $\text{DMSO}-d_6$) of compound 7.

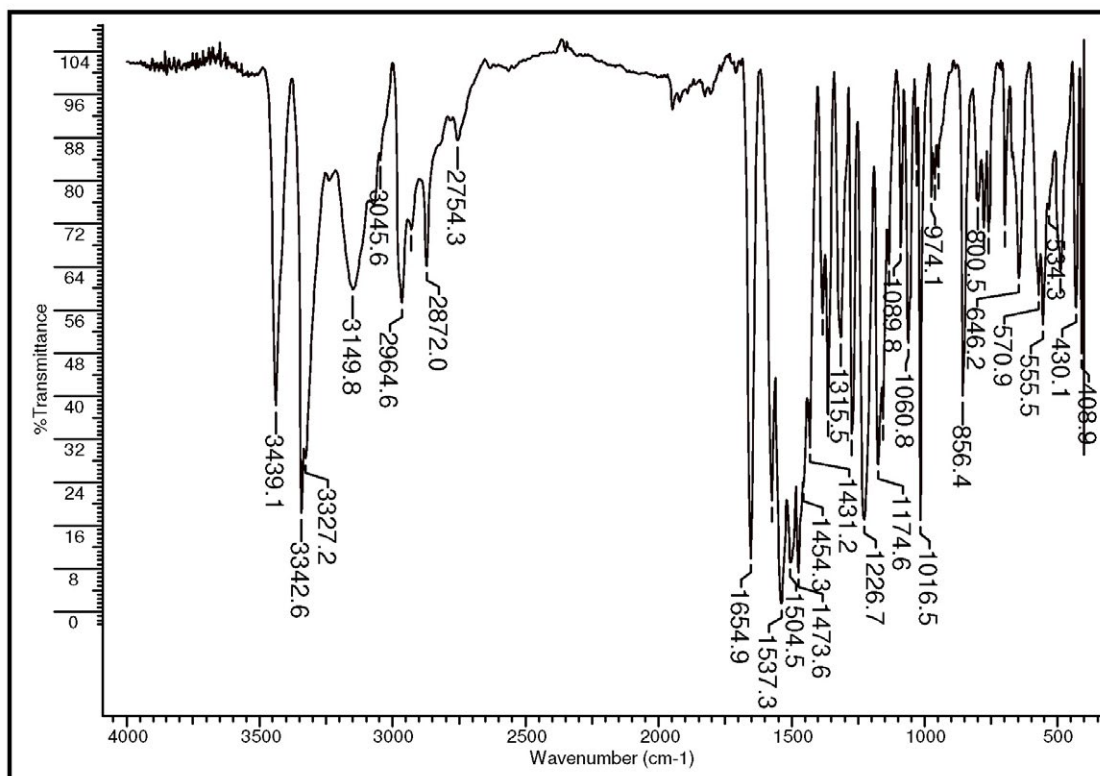


Figure S34. FTIR (KBr) spectrum of compound 8.

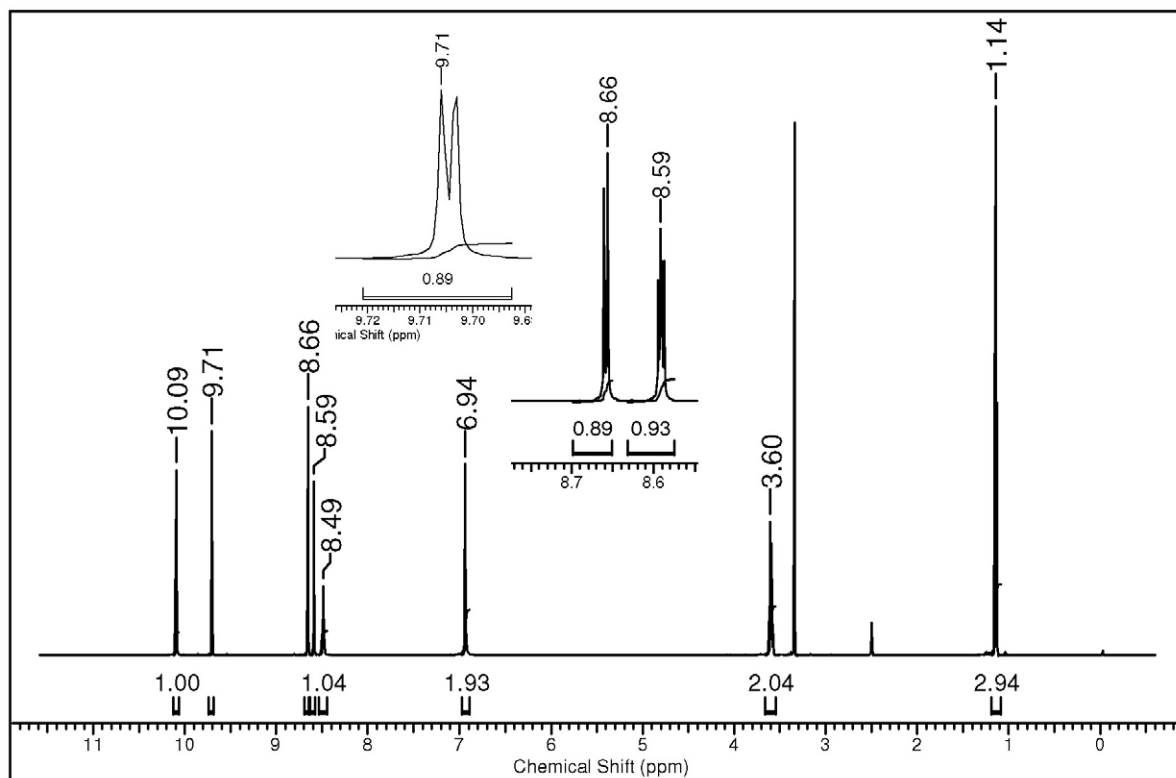


Figure S35. ¹H NMR spectrum (600 MHz, DMSO-*d*₆) of compound 8.

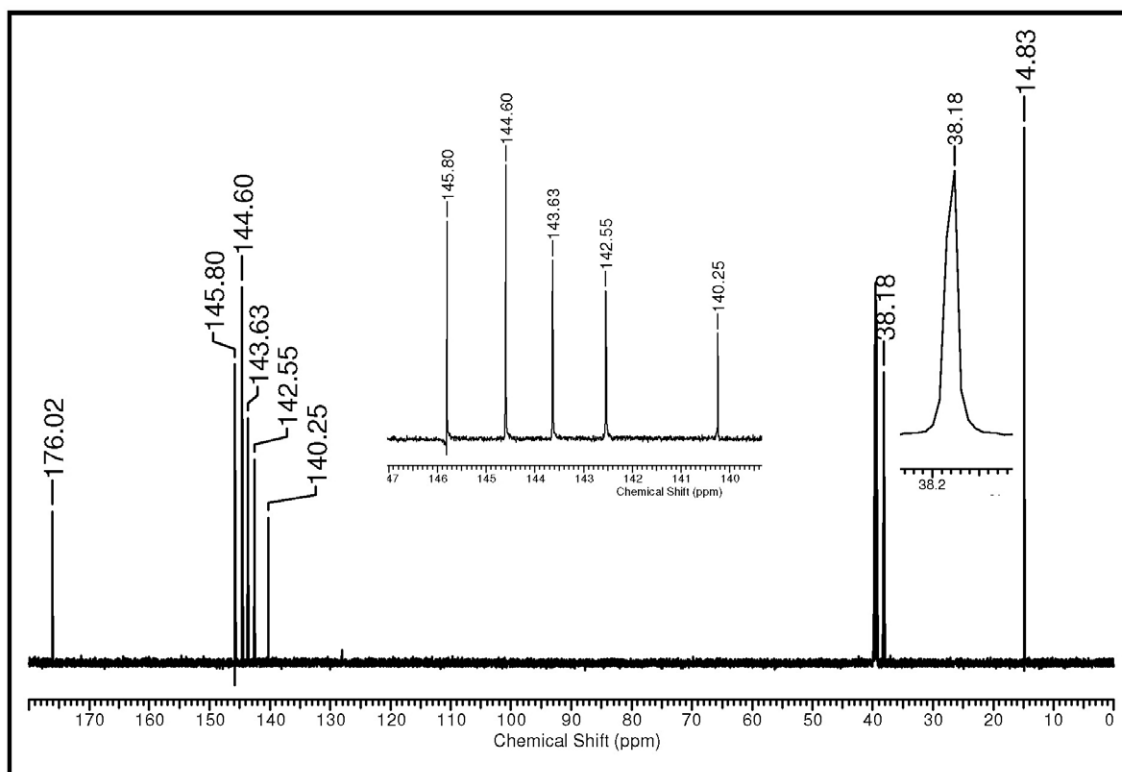


Figure S36. ^{13}C NMR spectrum (150 MHz, $\text{DMSO}-d_6$) of compound **8**.

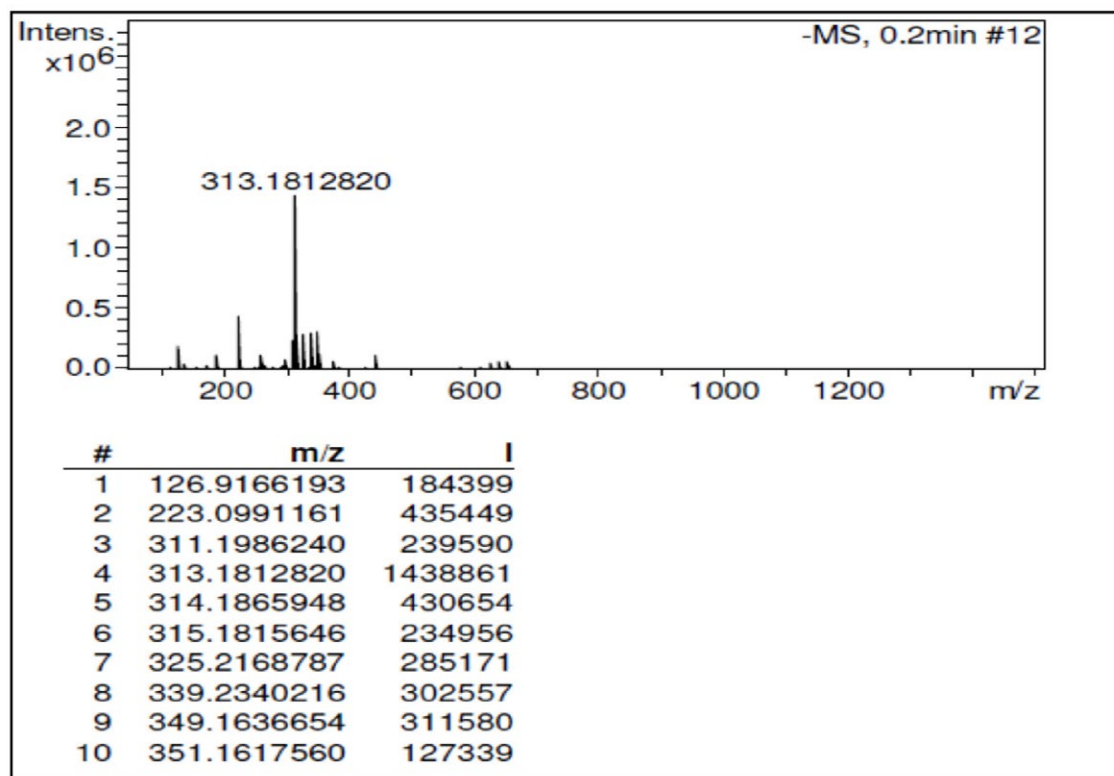


Figure S37. Mass spectrum (ESI-MS) of compound **8**.

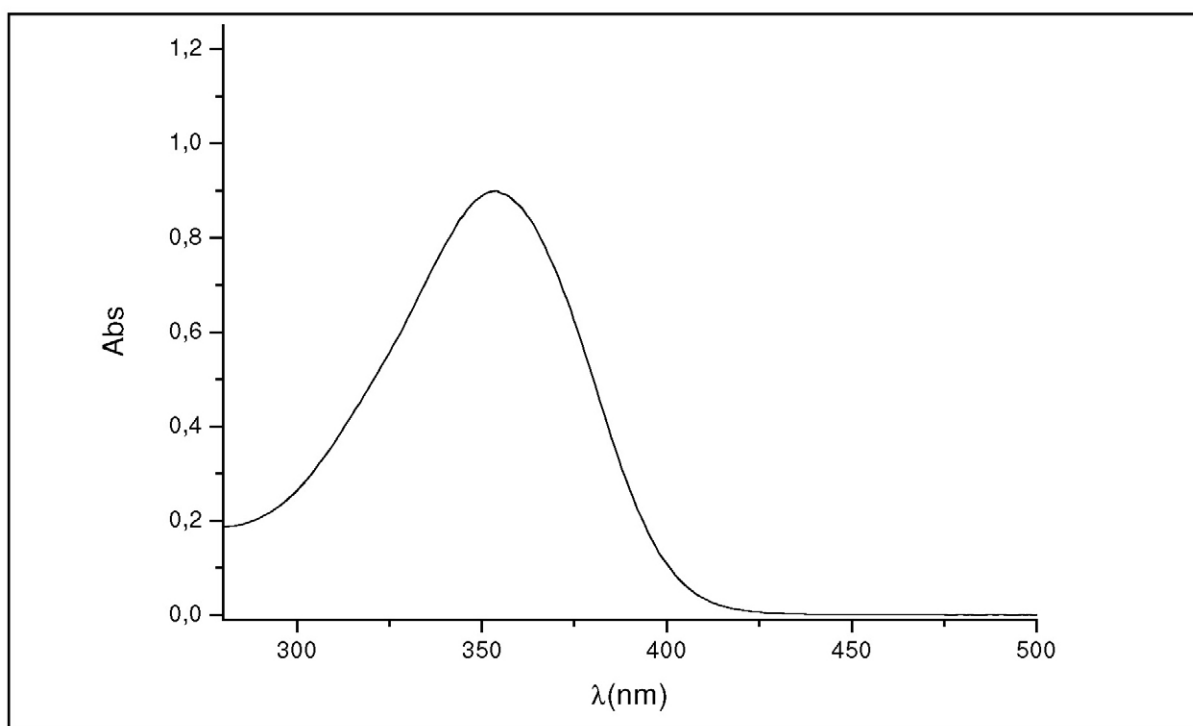


Figure S38. UV-Vis spectrum (DMF) of compound 8.

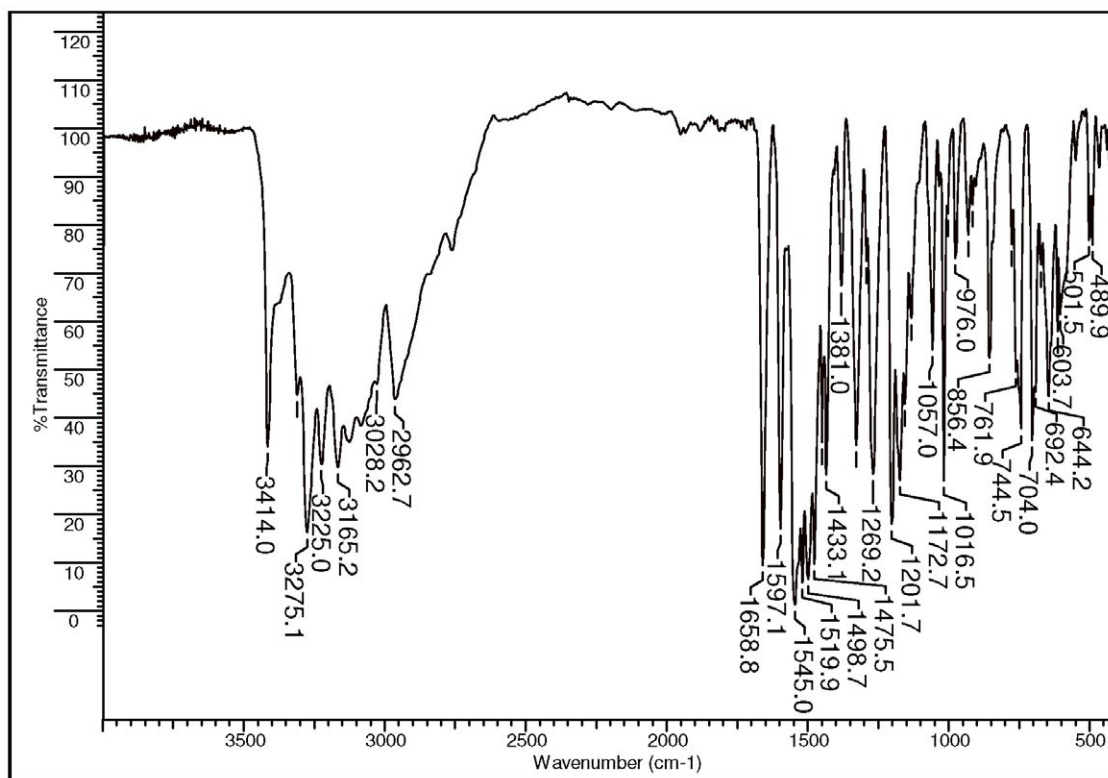


Figure S39. FTIR (KBr) spectrum of compound 9.

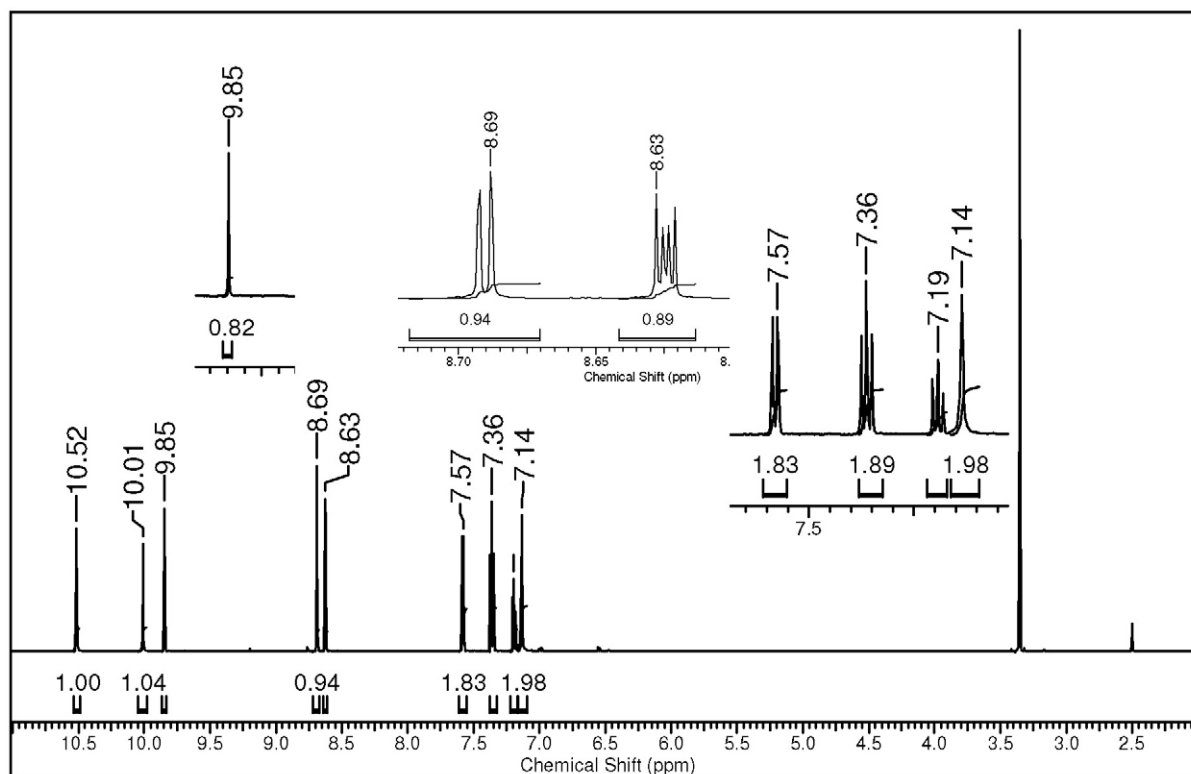


Figure S40. ¹H NMR spectrum (600 MHz, DMSO-*d*₆) of compound **9**.

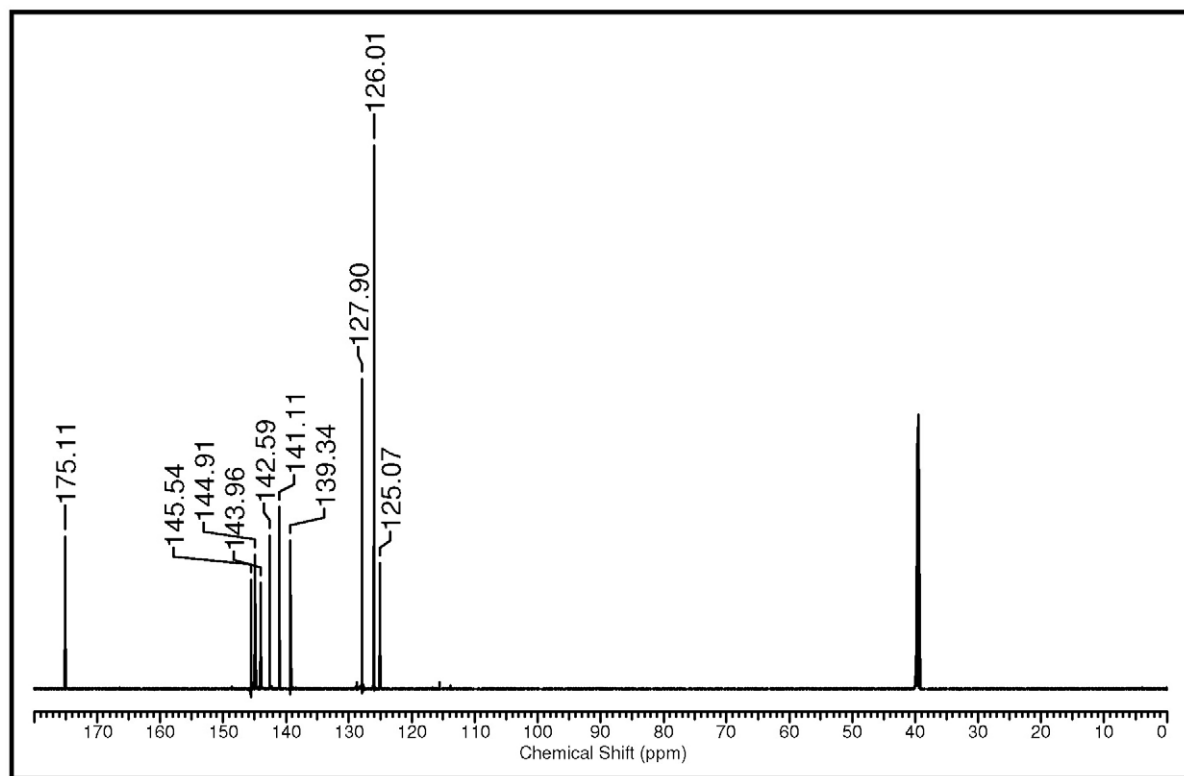


Figure S41. ¹³C NMR spectrum (150 MHz, DMSO-*d*₆) of compound **9**.

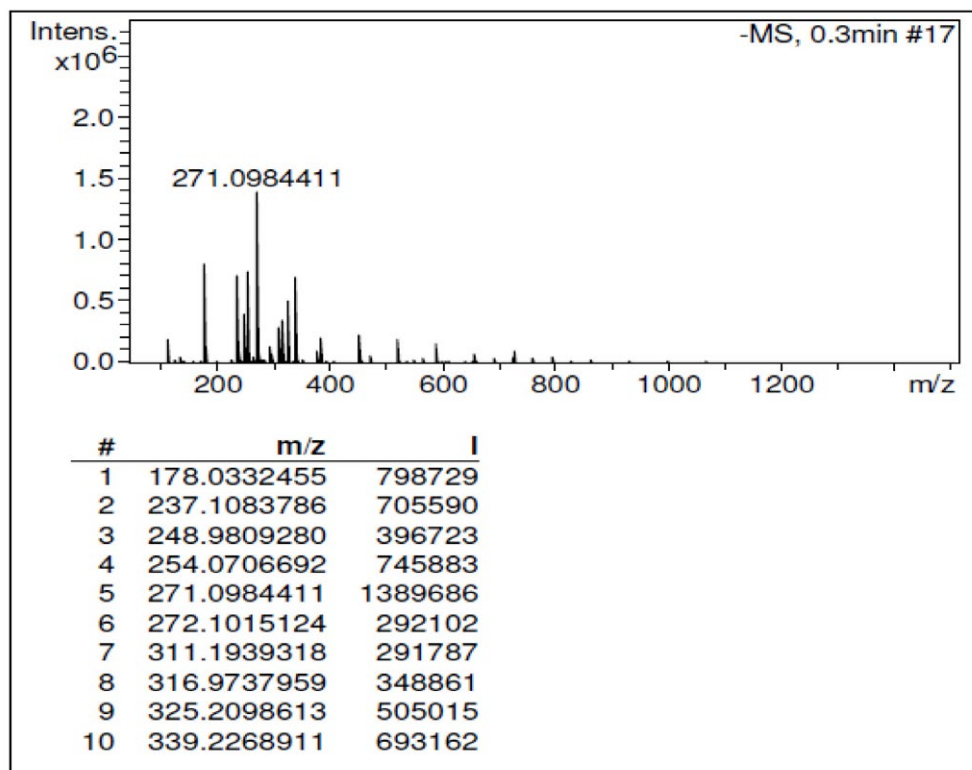


Figure S42. Mass spectrum (ESI-MS) of compound 9.

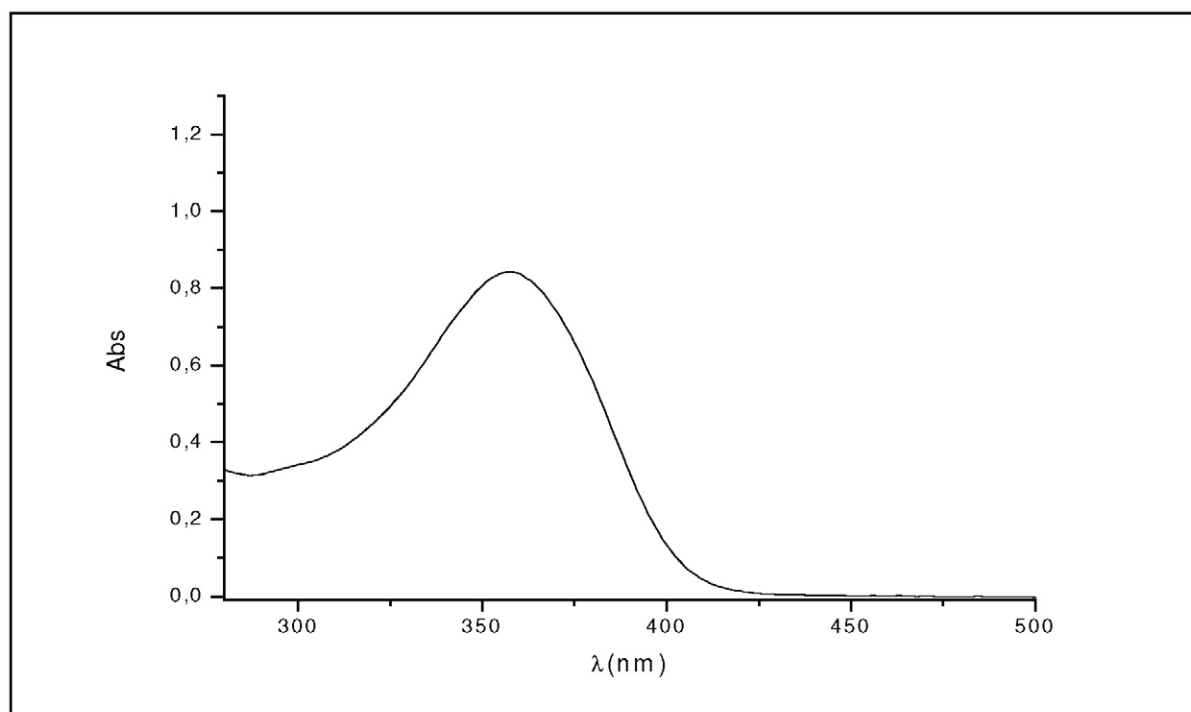


Figure S43. UV-Vis spectrum (DMF) of compound 9.

Table S1. Crystallographic data of [H₂PCIH]NO₃·H₂O (**1a**) and HPzAm4E (**8**)

Compound	[H ₂ PCIH]NO ₃ ·H ₂ O (1a)	HPzAm4E (8)
Empirical formula	C ₁₂ H ₁₃ N ₅ O ₅	C ₈ H ₁₂ N ₆ S
Formula weight	307.27	224.30
Temperature / K	293.15	296.92
Crystal system	triclinic	monoclinic
Space group	P-1	P2 ₁ /n
a ^a / Å	6.3407(4)	7.4889(3)
b ^a / Å	7.0760(6)	12.5626(5)
c ^a / Å	16.2140(12)	11.8649(4)
α ^a / degree	94.916(2)	90.00
β ^a / degree	99.164(2)	105.358(2)
γ ^a / degree	103.722(2)	90.00
Volume / Å ³	691.88(9)	1076.39(7)
Z ^b	2	4
ρ _{calc} ^c / (g cm ⁻³)	1.475	1.384
μ ^d / mm ⁻¹	0.118	0.278
F(000) ^e	320.0	472.0
Crystal size / mm ³	0.371 × 0.092 × 0.046	0.330 × 0.330 × 0.252
Radiation	Mo Kα (λ = 0.71073 Å)	Mo Kα (λ = 0.71073 Å)
2θ range for data collection / degree	5.14 to 50.68	4.82 to 52.3
Index ranges	-7 ≤ h ≤ 7, -8 ≤ k ≤ 8, -19 ≤ l ≤ 19	-9 ≤ h ≤ 9, -15 ≤ k ≤ 15, -14 ≤ l ≤ 14
Reflections collected	22129	16447
Independent reflections	2533 [R _{int} = 0.0378, R _{sigma} = 0.0184]	2144 [R _{int} = 0.0341, R _{sigma} = 0.0175]
Data/restraints/parameters	2533/0/199	2144/3/143
Goodness-of-fit on F ²	1.032	1.094
Final R indexes [I > 2σ (I)]	R ₁ = 0.0454, wR ₂ = 0.1088	R ₁ = 0.0365, wR ₂ = 0.0885
Final R indexes [all data]	R ₁ = 0.0622, wR ₂ = 0.1204	R ₁ = 0.0478, wR ₂ = 0.0972
Largest difference peak (hole) / (e Å ⁻³)	0.29 (-0.24)	0.24 (-0.28)

^aLattice parameters; ^bnumber of atoms *per* unit cell; ^ccalculated density; ^dabsorption coefficient; ^estructure factor evaluated in the zeroth-order case h = k = l = 0.

Table S2. Selected bond distances for [H₂PCIH]NO₃·H₂O (**1a**) and HPzAm4E (**8**) in comparison with HBPIH,² HPzAm4DH·H₂O³ and HPzAm4M·2H₂O³

	Bond distance / Å		
	HBPIH (3)	[H ₂ PCIH]NO ₃ ·H ₂ O (1a)	
C2–C7	1.4904(15)	1.459(3)	
C7–N2	1.2961(1)	1.272(2)	
N2–N3	1.3678(1)	1.375(2)	
N3–C8	1.3601(1)	1.345(2)	
C8–C9	1.5053(1)	1.496(3)	
C8–O1	1.2156(1)	1.220(2)	
	HPzAm4DH·H ₂ O	HPzAm4M·2H ₂ O	HPzAm4E (8)
C2–C7	1.479(6)	1.484(3)	1.483(2)
C7–N2	1.289(5)	1.295(2)	1.283(2)
N2–N3	1.337(5)	1.380(2)	1.3889(19)
N3–C8	1.355(5)	1.350(3)	1.359(2)
C8–S1	1.690(5)	1.696(2)	1.6765(17)
C8–N4	1.316(6)	1.326(2)	1.323(2)
N4–C9	–	–	1.456(2)
C7–N5	1.335(6)	1.341(2)	1.359(2)

Table S3. Selected angles for [H₂PClH]NO₃·H₂O (**1a**) and HPzAm4E (**8**) in comparison with HBPIH,² HPzAm4DH·H₂O³ and HPzAm4M·2H₂O³

	Bond angle / degree		
	HBPIH	[H ₂ PClH]NO ₃ ·H ₂ O (1a)	
N1–C2–C7	118.05(9)	114.18(17)	
C2–C7–N2	127.53(9)	121.57(18)	
C7–N2–N3	117.83(9)	115.20(16)	
N2–N3–C8	120.29(9)	118.86(15)	
N3–C8–O1	124.84(1)	124.51(19)	
N3–C8–C9	113.26(9)	115.26(16)	
	HPzAm4DH·H ₂ O	HPzAm4M·2H ₂ O	HPzAm4E (8)
N1–C2–C7	118.6(4)	117.1(2)	116.79(15)
C2–C7–N2	113.8(4)	115.7(2)	116.54(14)
C7–N2–N3	119.0(4)	116.8(2)	117.19(14)
N2–N3–C8	117.4(4)	118.5(2)	116.93(14)
N3–C8–S1	120.6(4)	120.1(2)	120.73(13)
N3–C8–N4	115.3(5)	116.8(2)	115.62(15)
C8–N4–C9	–	–	125.40(16)

Table S4. Selected dihedral angles for [H₂PClH]NO₃·H₂O (**1a**) and HPzAm4E (**8**)

	Dihedral angle / degree	
	[H ₂ PClH]NO ₃ ·H ₂ O (1a)	HPzAm4E (8)
N1–C2–C7–N2	178.83(18)	177.20(16)
C2–C7–N2–N3	179.73(17)	177.87(14)
C7–N2–N3–C8	178.14(18)	171.71(16)
N2–N3–C8–O1	–1.8(3)	–
N2–N3–C8–S1	–	173.94(12)

Table S5. Hydrogen bond distances and angles for [H₂PClH]NO₃·H₂O (**1a**) and HPzAm4E (**8**)

Compound	D–H···A	D–H / Å	H···A / Å	D···A / Å	D–H···A / degree
1a	N4–H4···O11 ^a	0.86	2.07	2.872(3)	154
	N4–H4···O13 ^a	0.86	2.09	2.835(3)	145
	N4–H4···N1 ^a	0.86	2.46	3.308(3)	170
	N3–H3···O1W ^b	0.86	1.96	2.799(2)	166
	O1W–H1WA···N1 ^c	0.93	2.03	2.917(2)	158
	O1W–H1WB···O1	0.90	2.04	2.859(2)	150
	O1W–H1WB···N2	0.90	2.54	3.220(2)	132
8	N4–H4···N2	0.86	2.15	2.564(2)	109
	N3–H3···N6 ^d	0.86	2.34	3.119(2)	151
	N5–H5A···N1	0.85	2.28	2.692(2)	110(2)
	N5–H5B···S1 ^e	0.85	2.70	3.364(2)	136(2)

Symmetry codes: ^a–x + 2, –y + 2, –z + 1; ^bx + 1, y, z; ^c–x + 1, –y + 2, –z (compound **1a**); ^d–x + 1/2, y – 1/2, –z + 3/2; ^e–x, –y + 1, –z + 1 (compound **8**).

References

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